

# Pumping stations and water transport

Water quality aspects of drinking water  
networks

ct5550

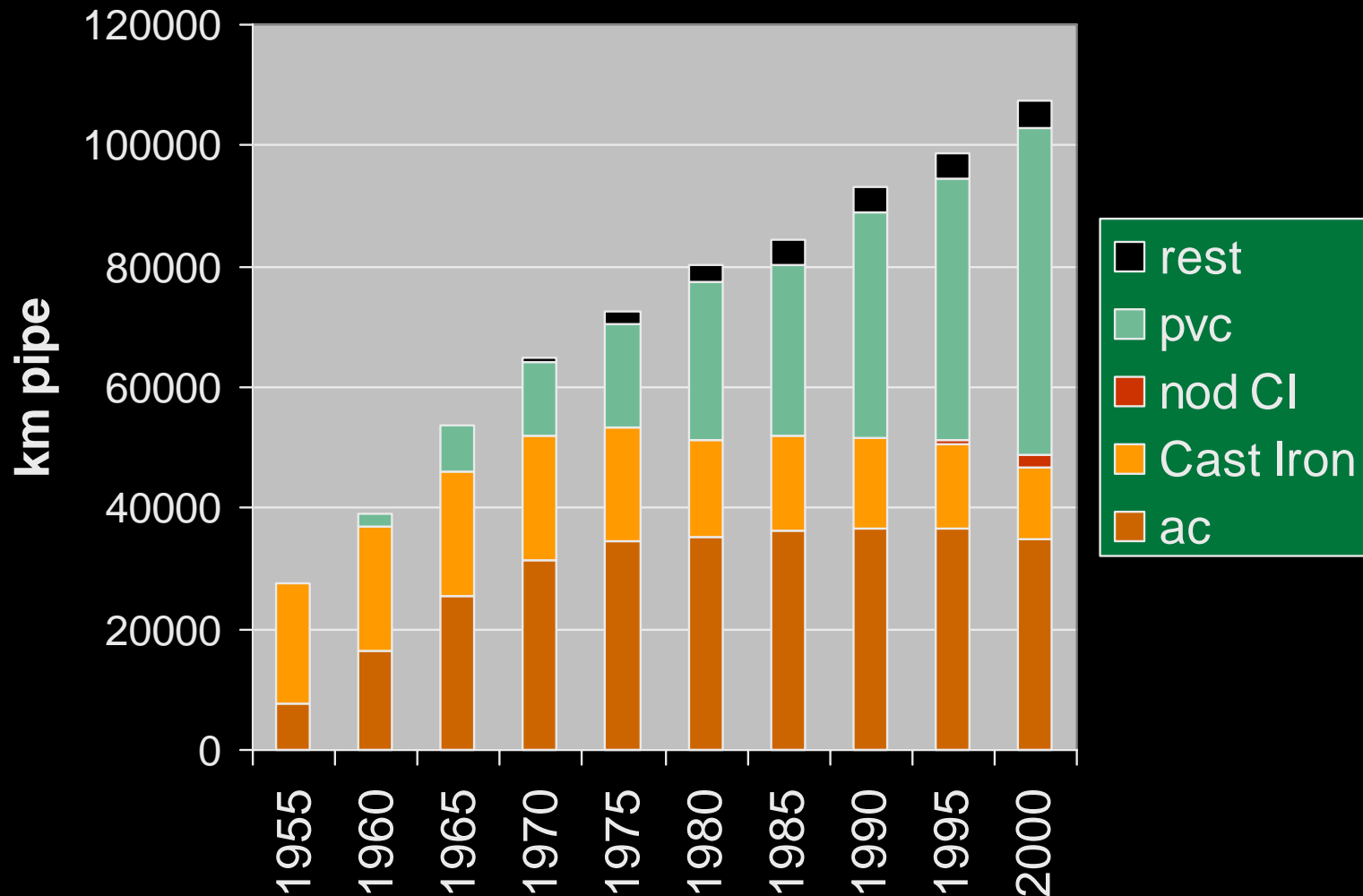
A thick, horizontal yellow brushstroke with a textured, painterly appearance, spanning most of the width of the slide.

# Introduction

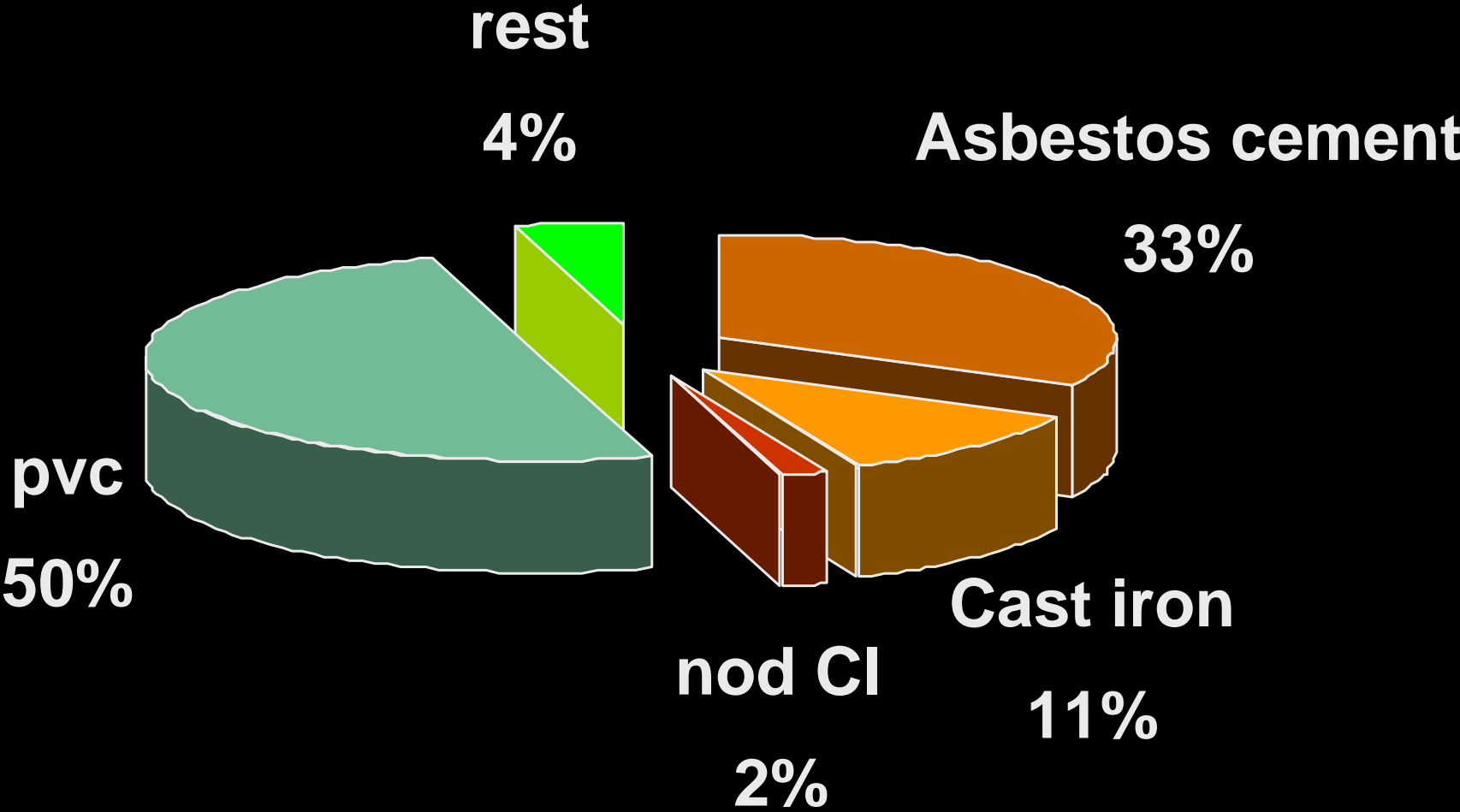
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- Water quality changes during transport through the network
- Chemical processes: interaction between pipe material and water
- Biological processes: development/regrowth of bacteria and invertebrates
- Physical processes: sedimentation and resuspension

# Material composition network



# Present material composition



# Physical processes

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- Main problem: discoloured water
- Primarily a customer related problem: discoloured water is 'harmless'
  - Although relation with bacteriological problems
- Focus because of changing attitude of both customer (more critical and outspoken) as water company (more customer oriented)

# Early 90's: red water on the agenda in the Netherlands

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- Increased customer complaints
  - more critical attitude
  - more professional registration
- Unexpected
  - decrease in cast iron
  - also in new plastic networks
- Unsatisfactory results of (costly) cleaning programs

# Ø100 Cast Iron (1900)



# Basic questions

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- What is discoloured water
- How to measure discoloured water
- What is the cause and nature of discoloured water
- What can we do about it

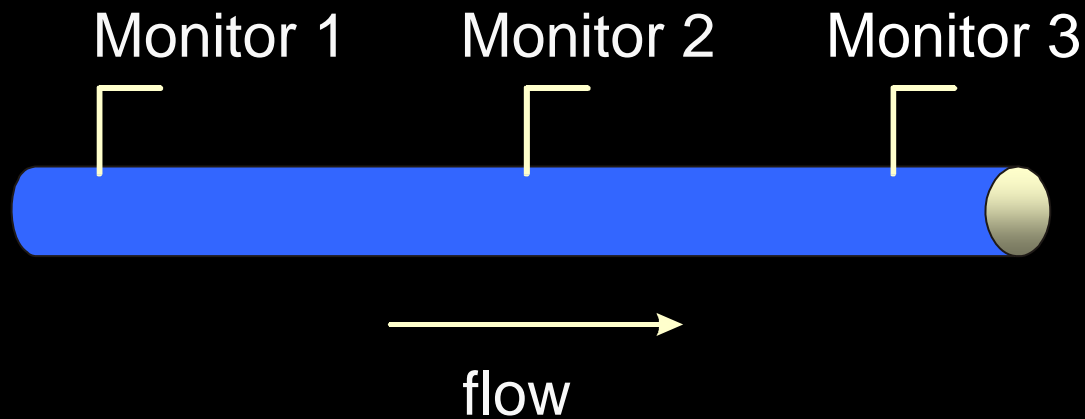


# What is red water?

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- Water with a turbidity that can be noticed by a customer
- Turbidity is the key factor in red water

# How to measure red water: Monitoring System



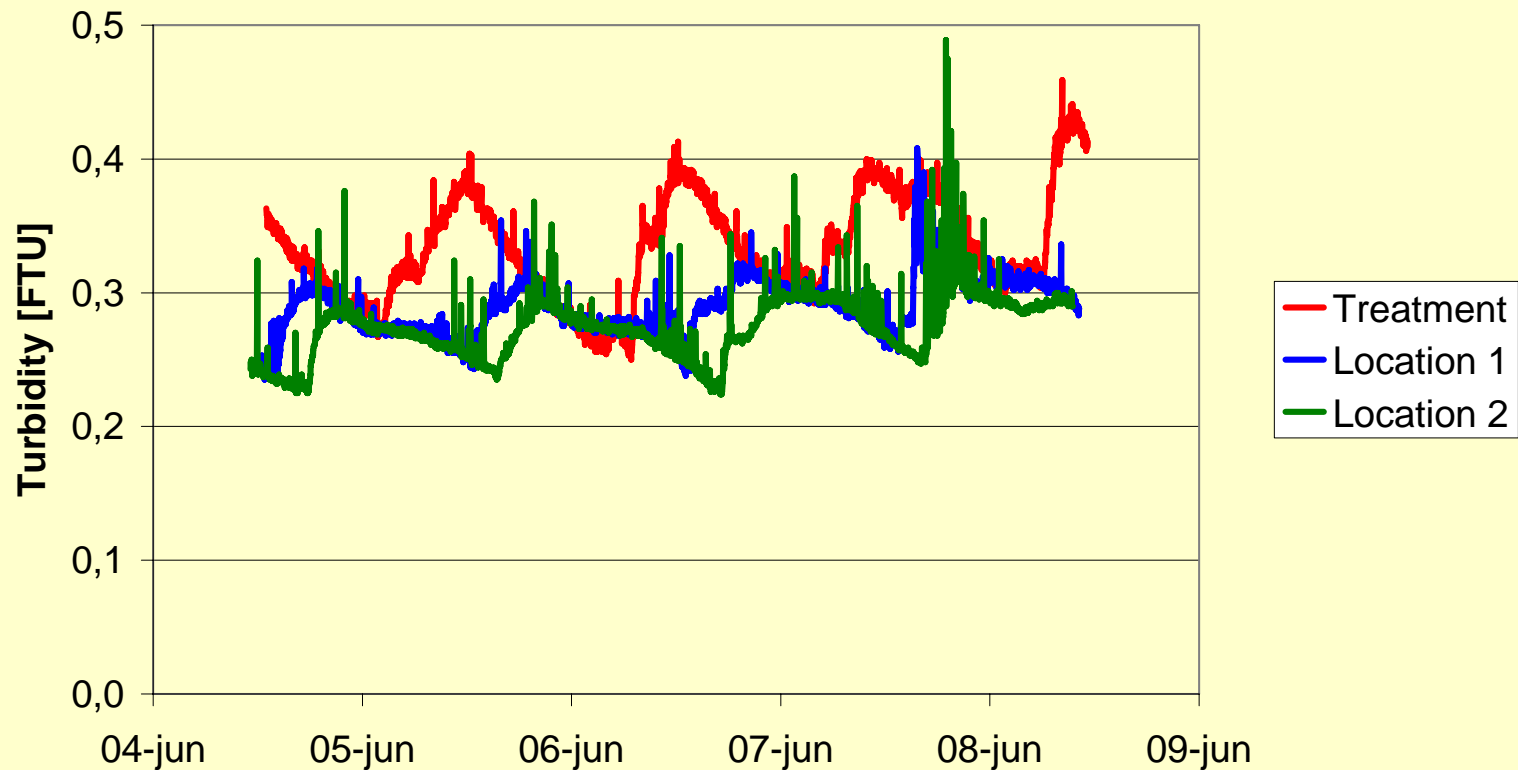
- pH
- Oxygen
- Turbidity
- Temperature
- Conductivity
- Pressure
- (Velocity)

# Single Monitoring system



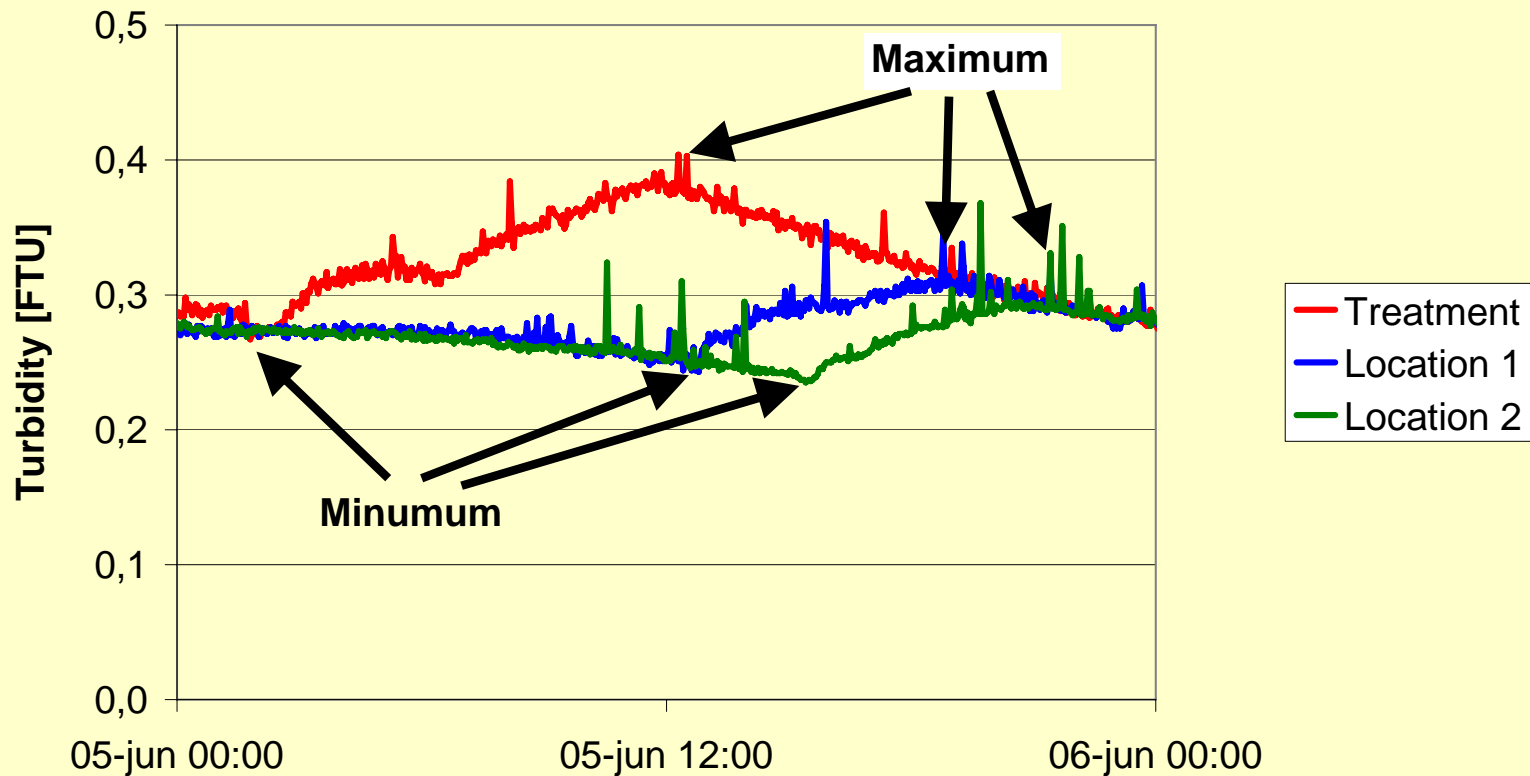
# Results monitoring turbidity

## Continuous monitoring turbidity

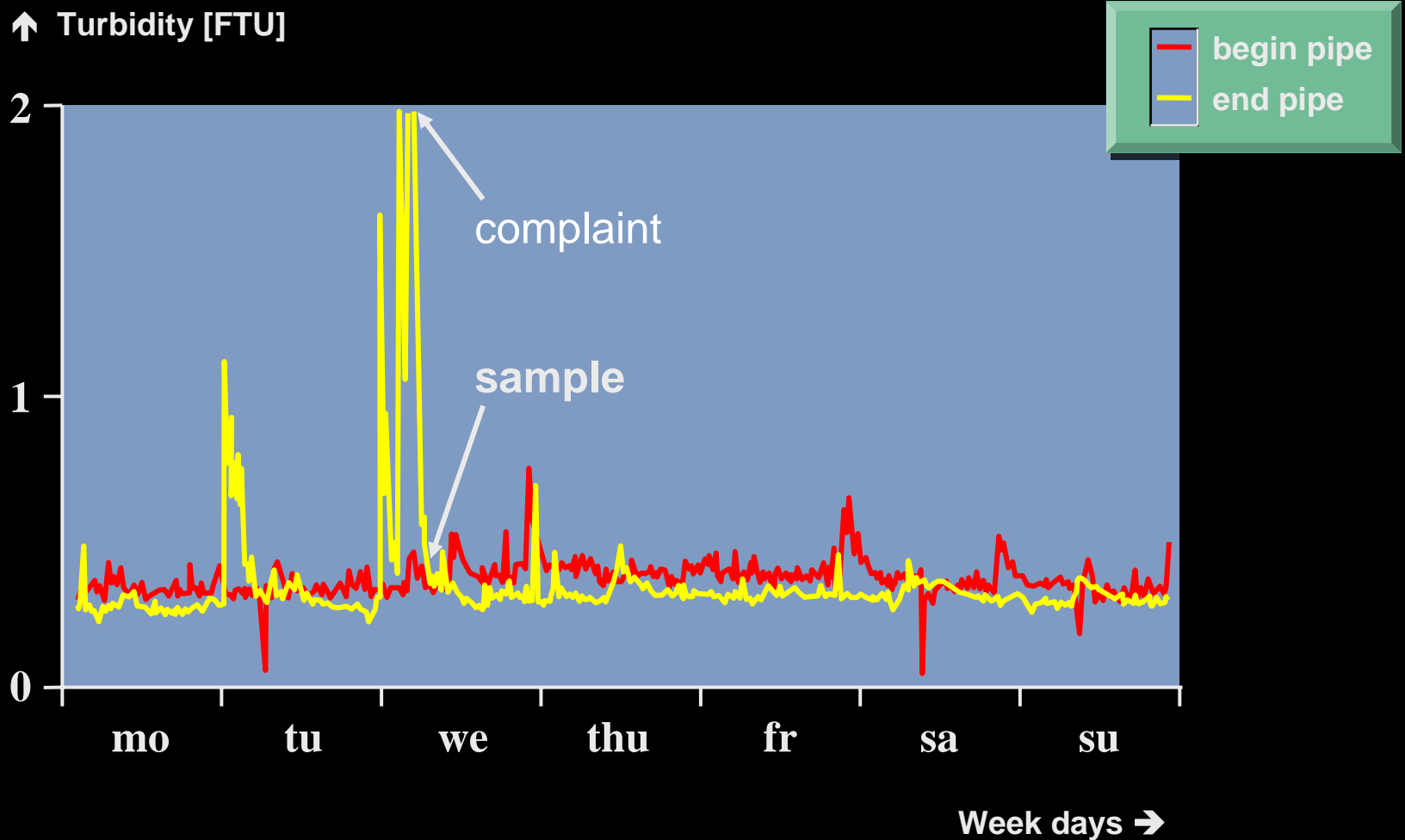


# Closer detail

## Continuous monitoring turbidity



# Continuous monitoring of turbidity

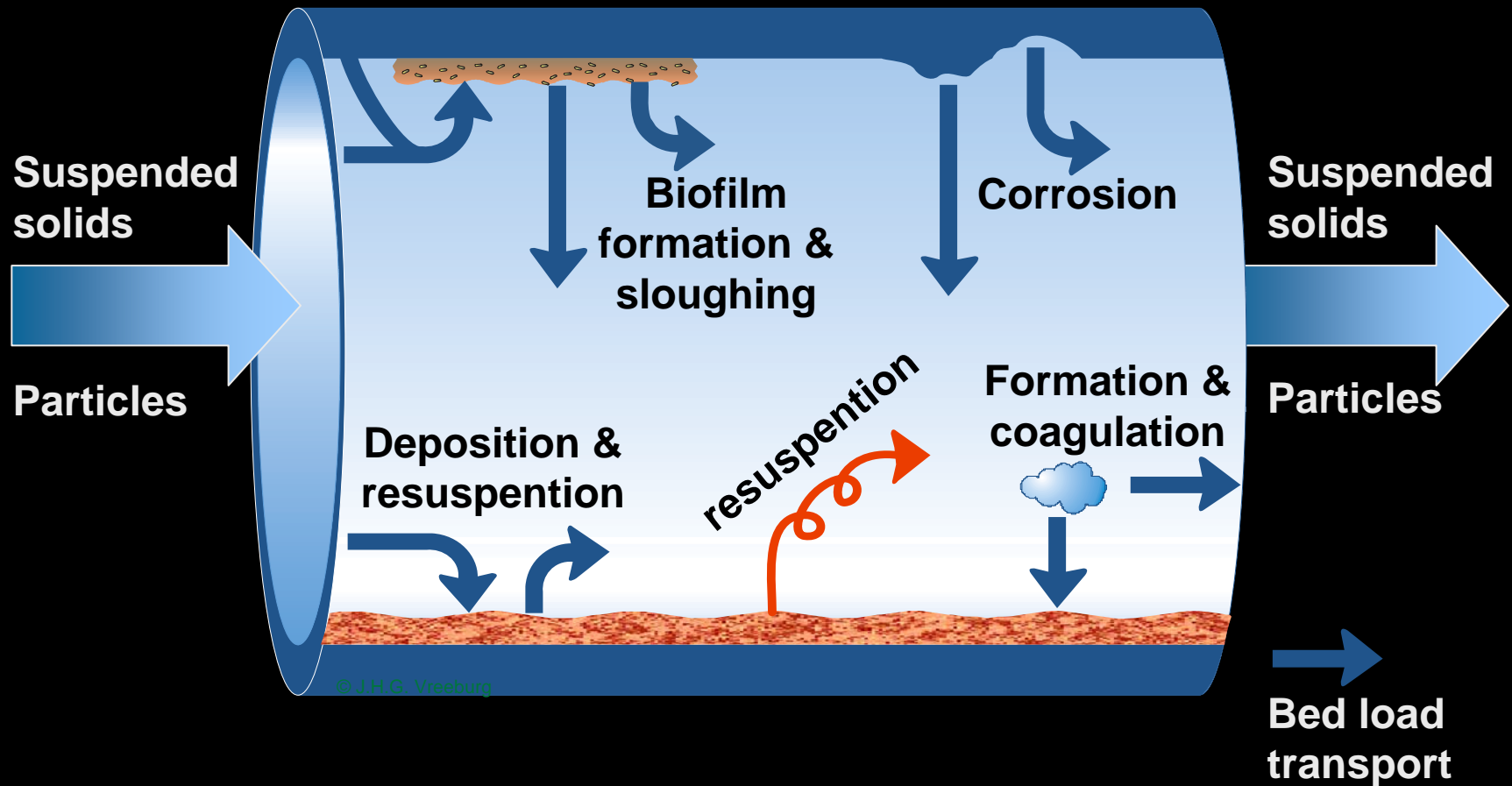


# Cause of discoloured water

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- Resuspension of sediment
- Sediment originates from
  - source/treatment plant
  - network (corroding cast iron, biofilm)
  - mixing different water types
- Sedimentation is promoted by low velocities in the network
- Sudden changes in velocity promote resuspension

# Mass balance in a network



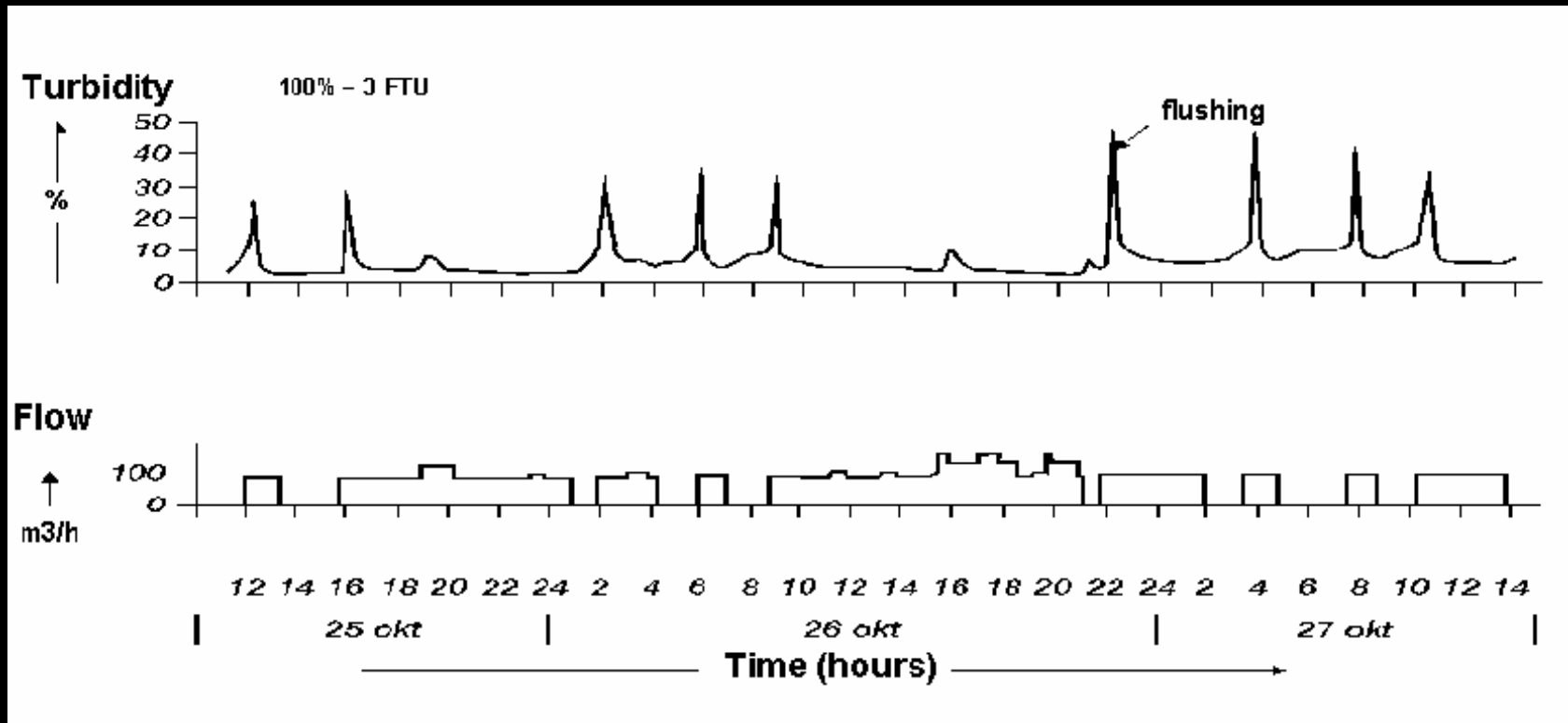


# Actions to prevent red water: a three stage rocket

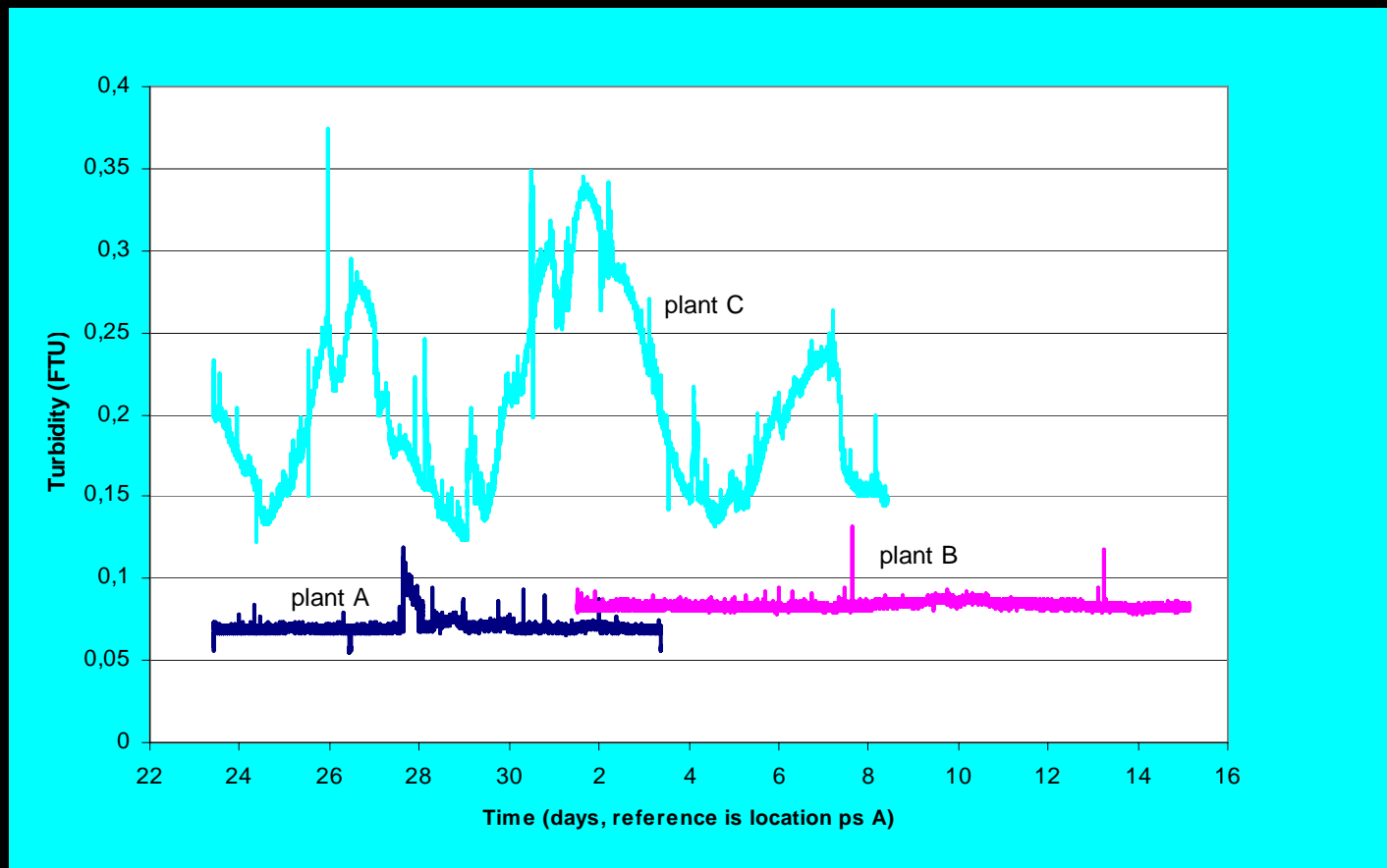
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- Prevent the sediment from entering the network
  - Improve the treatment of water
- Remove sediment swiftly en effectively
  - Conduct cleaning programmes
- Prevent sedimentation of suspended matter
  - Design small diameter pipes: design for self-cleaning capacity

# Variation of production capacity and water quality



# Turbidity different pumping stations



# First stage: improve treatment

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- Goal: steady process with low turbidity
- Tools: demand forecasting and clever use of balancing reservoirs
- But: conventional treatment leaves sediment in the water

# Second stage: Remove sediment

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- Cleaning the network, available techniques
  - Flushing
  - Water/air scouring
  - Pigging
- Measuring method:
  - Resuspension potential

# Research tools

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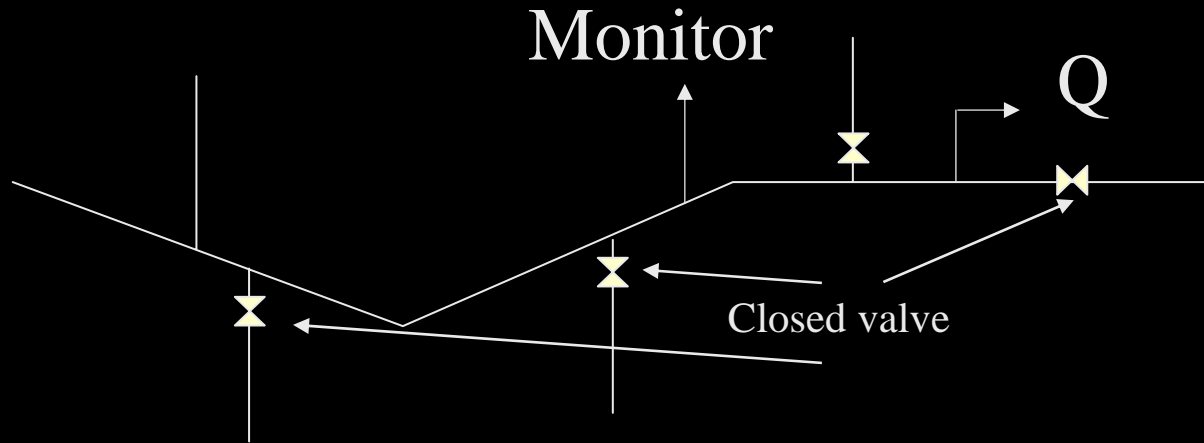
- Continuous monitoring turbidity and Particle counts
- Resuspension Potential Method
- Experimental pipe test rig
- Analysis of concentrate MF-installation

# Resuspension Potential Method

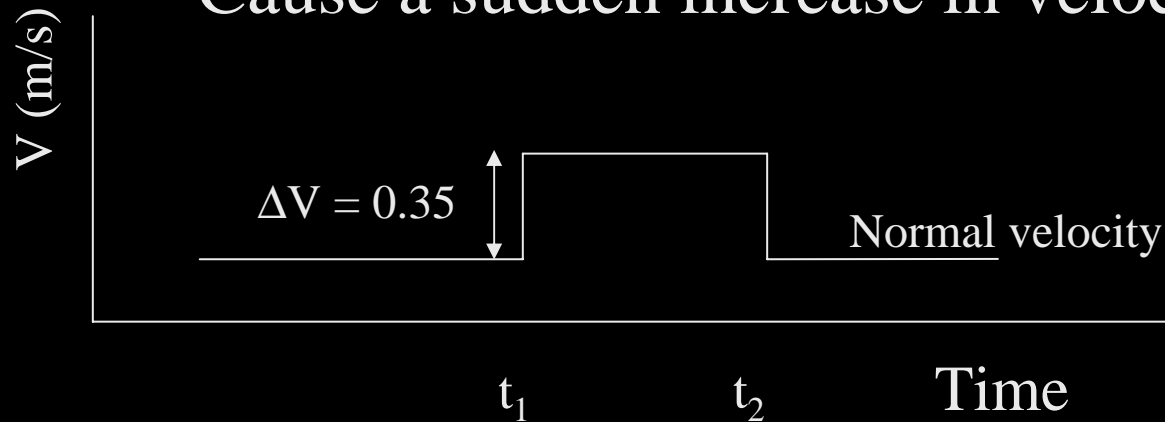
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- Principle
  - resuspension of sediment by sudden increase of velocity
  - 0,35 m/s on top of normal velocity
  - turbidity monitoring
  - no impact on the customer

# Resuspension Potential Method



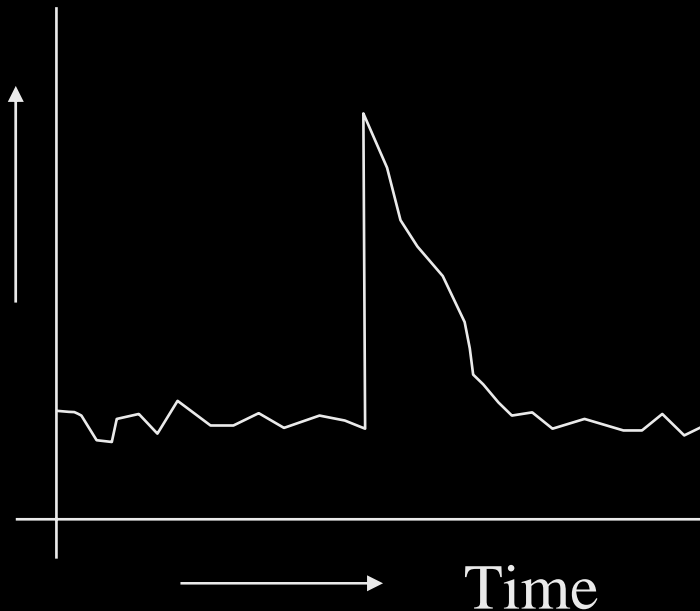
Cause a sudden increase in velocity





# Typical RP curve

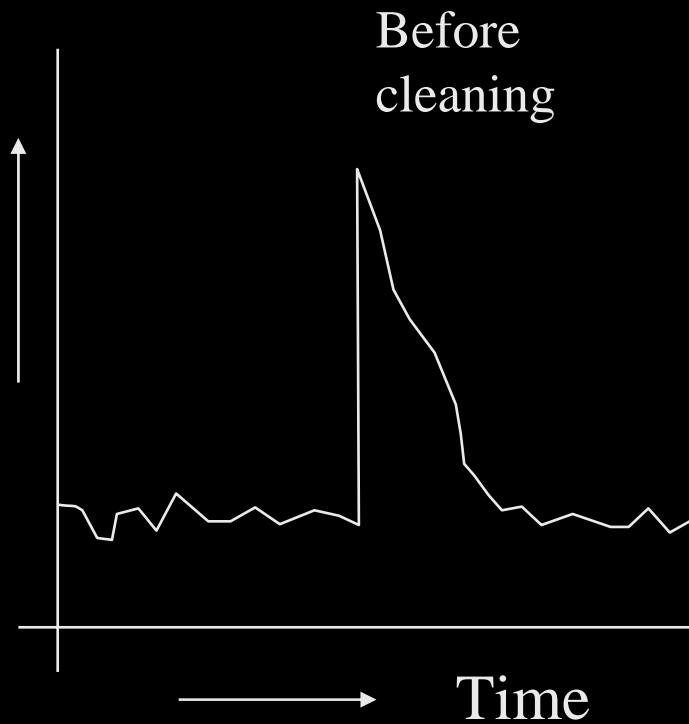
Turbidity



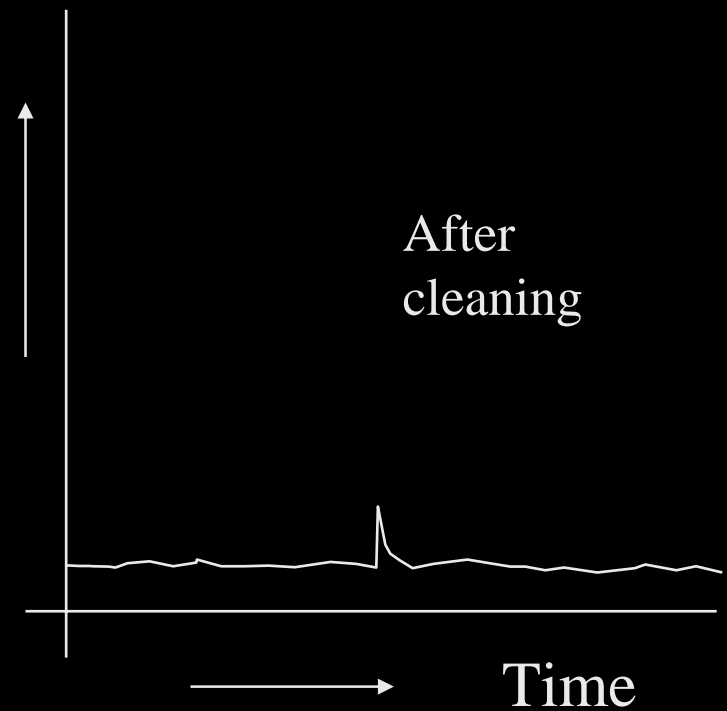
High resuspension potential indicates polluted network: cleaning network, e.g. flushing

# Effect of flushing monitored with resuspension potential

Turbidity



Turbidity



# Pictures monitor connection



# Resuspension Potential Method



# Resuspension Potential Method



# Resuspension potential method

## Sample measuring



# Typical RPM curve



# Relative interpretation RPM

- 5 aspects determine the RP:
  - Absolute maximum turbidity first five minutes
  - Average turbidity first five minutes
  - Absolute maximum turbidity last ten minutes
  - Average turbidity last ten minutes
  - Resetling time
- 0 to 3 'points' per aspect



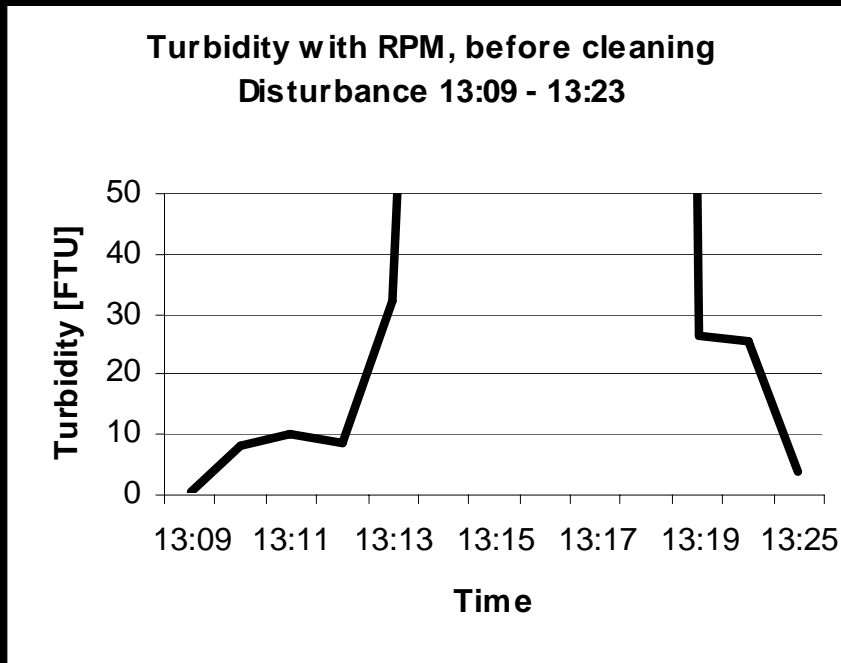


# Ranking RPM

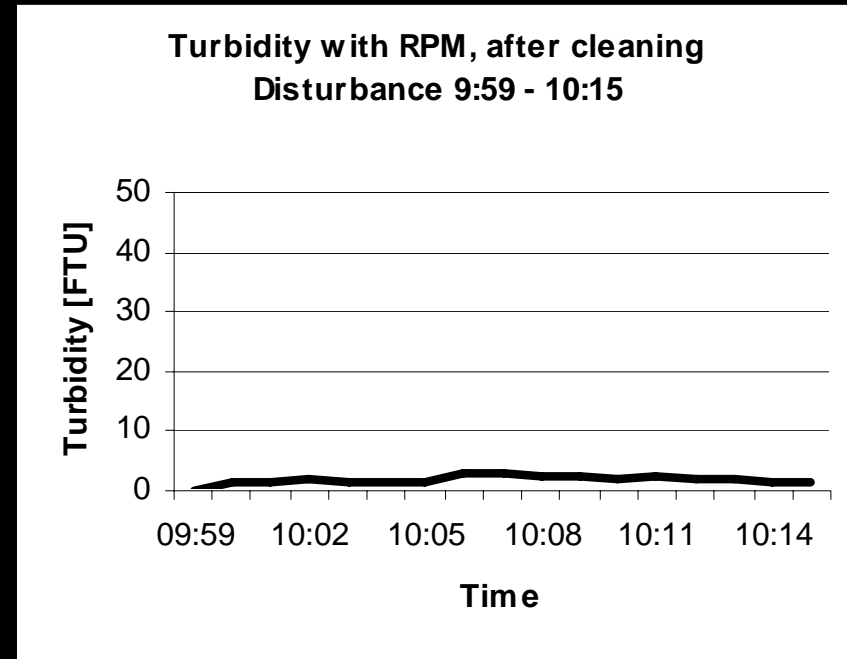
## Turbidity Dr Lange, Hydrant measuring

Category \ points	0	1	2	3
Absolute max first 5 min	<3 ftu	3 –10 ftu	10-40 ftu	>40 ftu
Average first 5 min	<3 ftu	3 –10 ftu	10-40 ftu	>40 ftu
Absolute max last 10 min	<3 ftu	3 –10 ftu	10-40 ftu	>40 ftu
Average max last 10 min	<3 ftu	3 –10 ftu	10-40 ftu	>40 ftu
Time to clear	< 5 min.	5-15 min	15-60 min	>60 min

# Typical RPM results Pre and Post cleaning

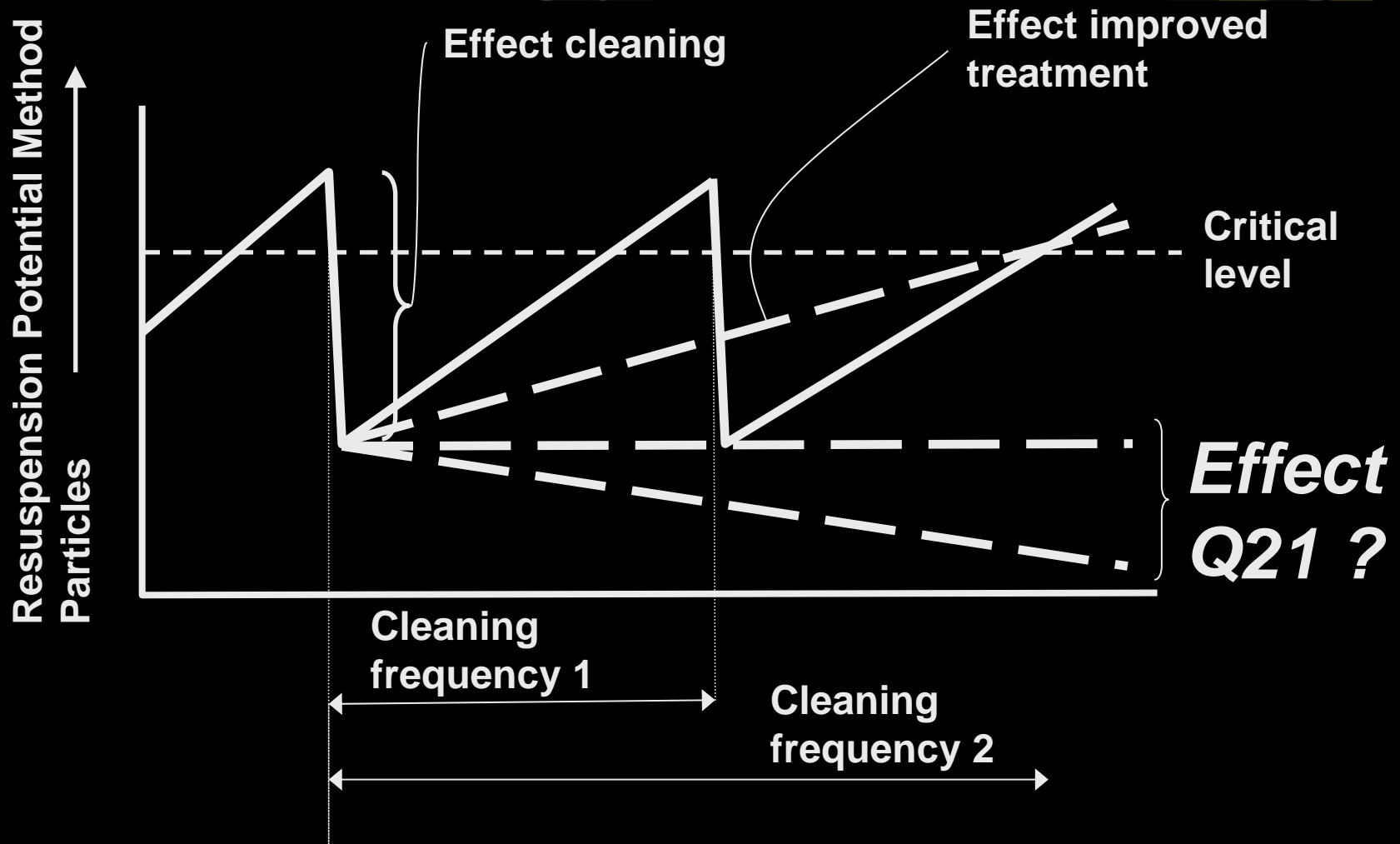


Score 12



Score 0

# Hypothesis effects improved particle load



# Additional research

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- Particle load present system
  - Large volume sampling
  - Test rig Ø100
  - “new” test rig multiple pipes
- Goals:
  - Determine characteristics present drinking water
  - Develop new measurement methods and tools
  - Trigger students

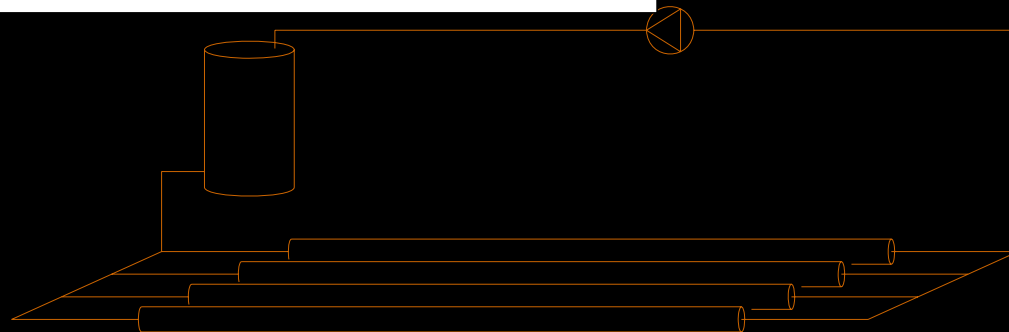
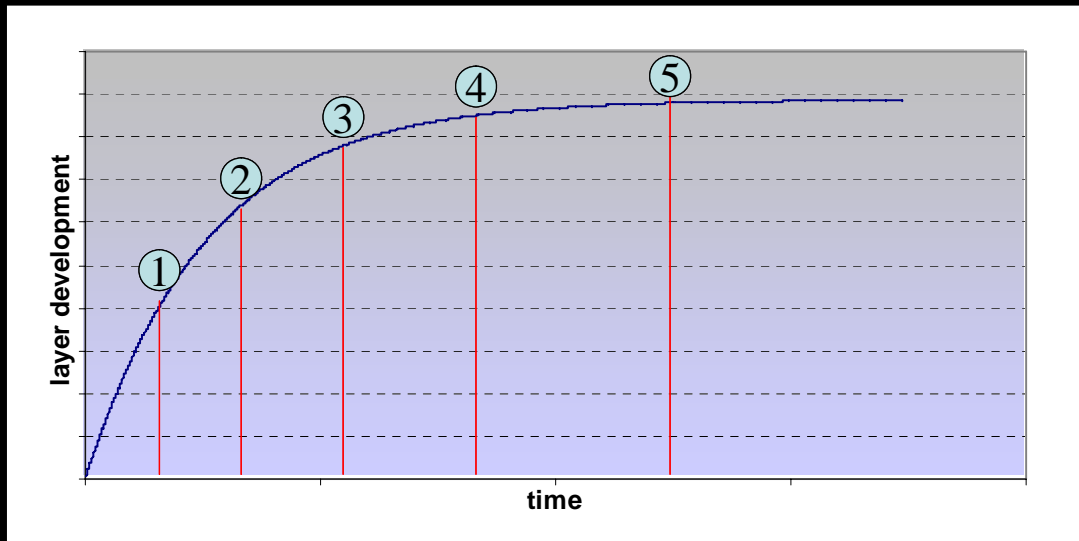
# Test pipe rig Ø100



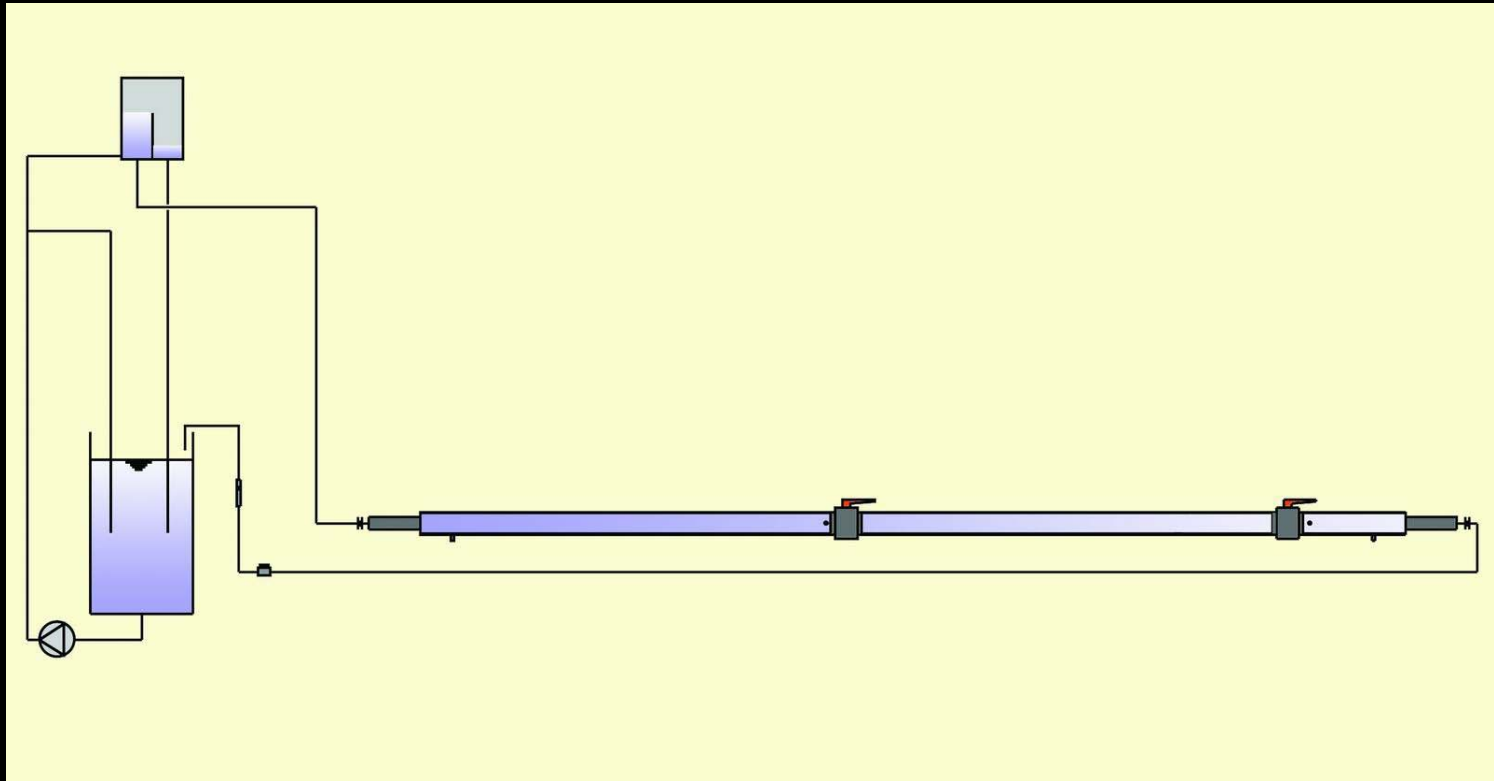
**Charge with  
concentrate**



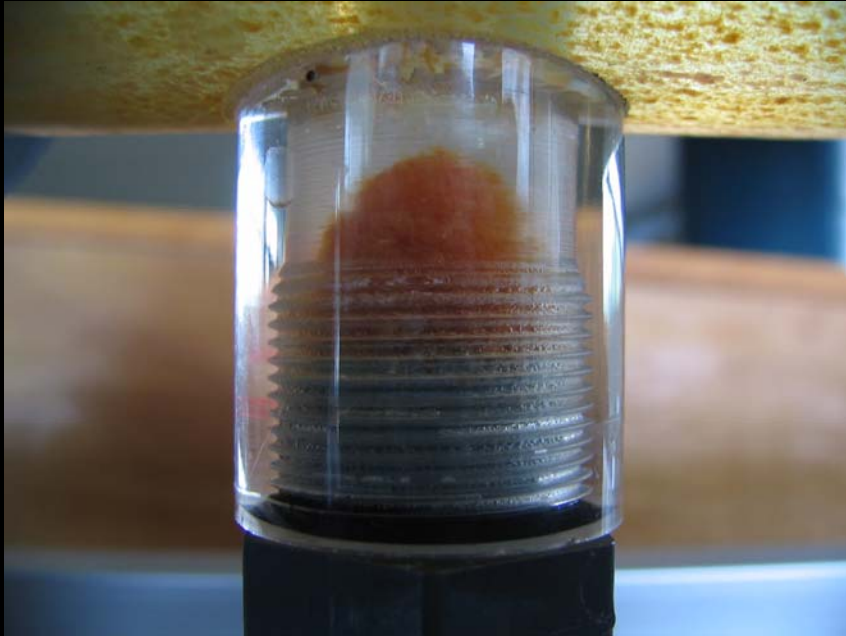
# Original set up test rig



# Practical set up

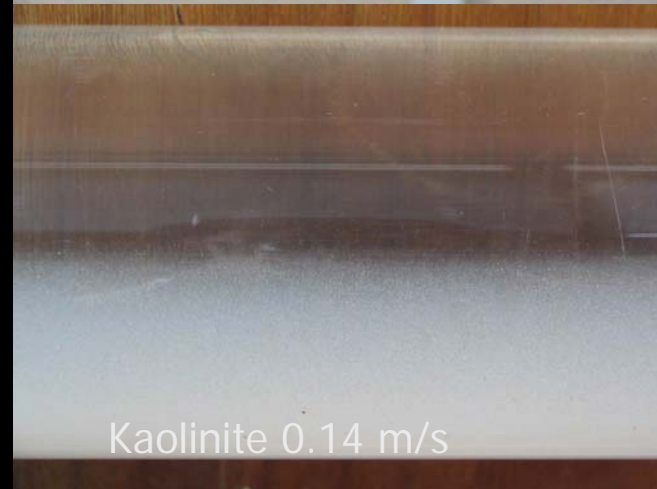
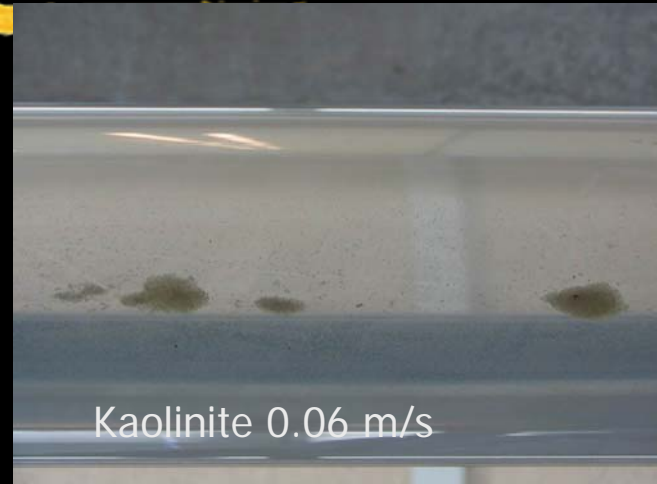
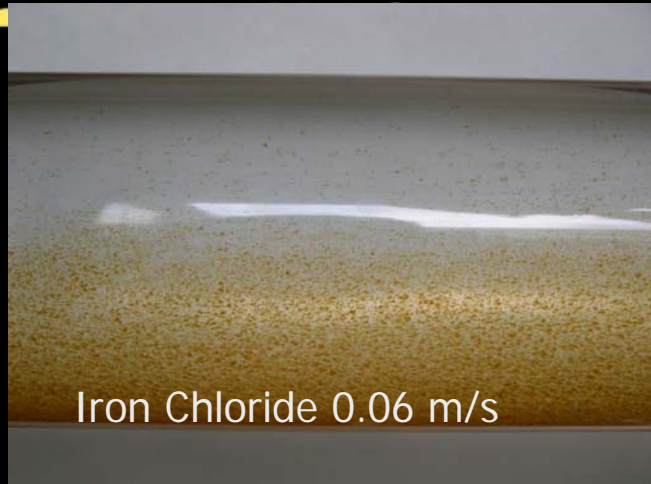


# First observations





# Some more observations



# Drinking water quality aspects: Remove the sediment

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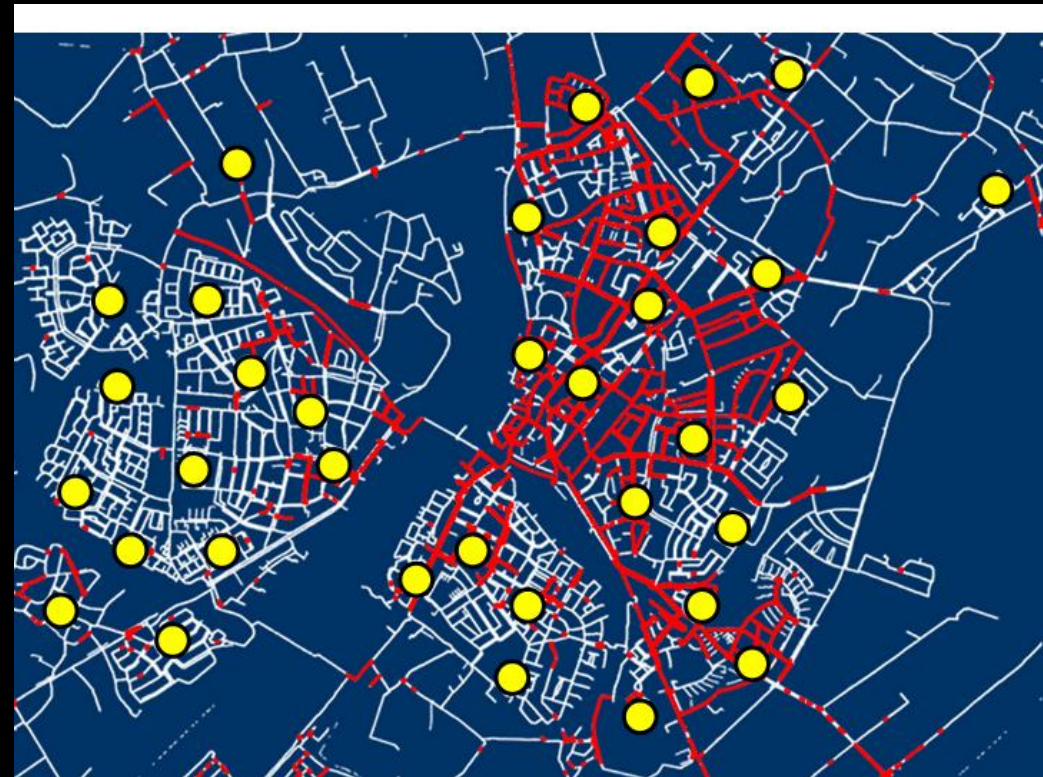
# Remove the sediment

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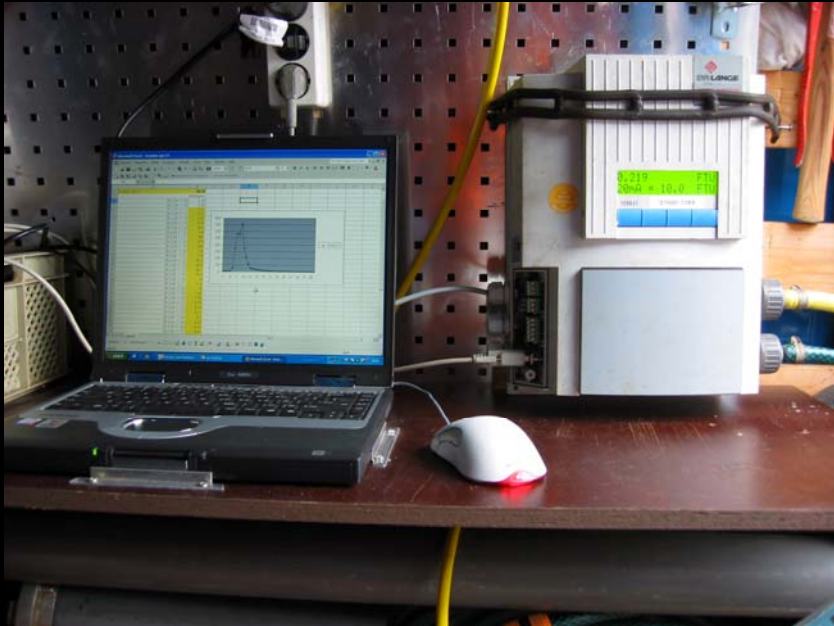
- Water flushing is effective if:
  - 1,5 m/s is reached
  - Flush volume is 3 times the pipe content
  - Clear water front
- Alternative methods as water/air scouring and pigging only when water flushing boundaries are not met

# Results of cleaning program

- City of Venlo
- Red line are CI
- Rest is PVC/AC

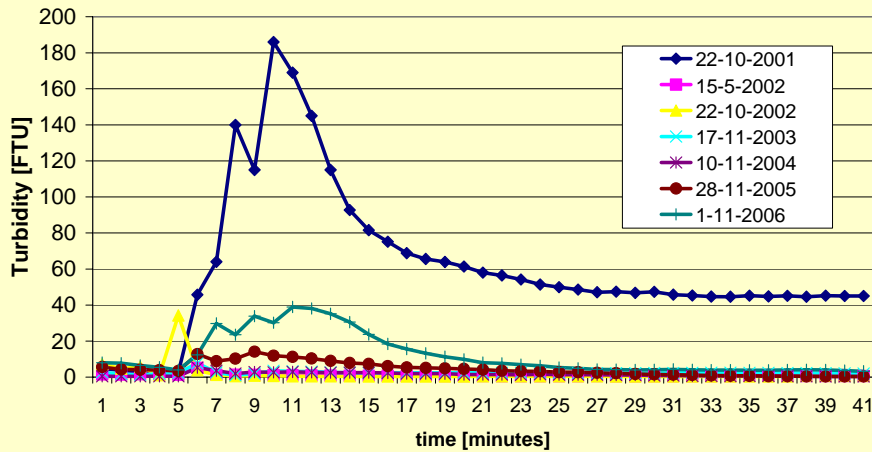


# Measuring equipment

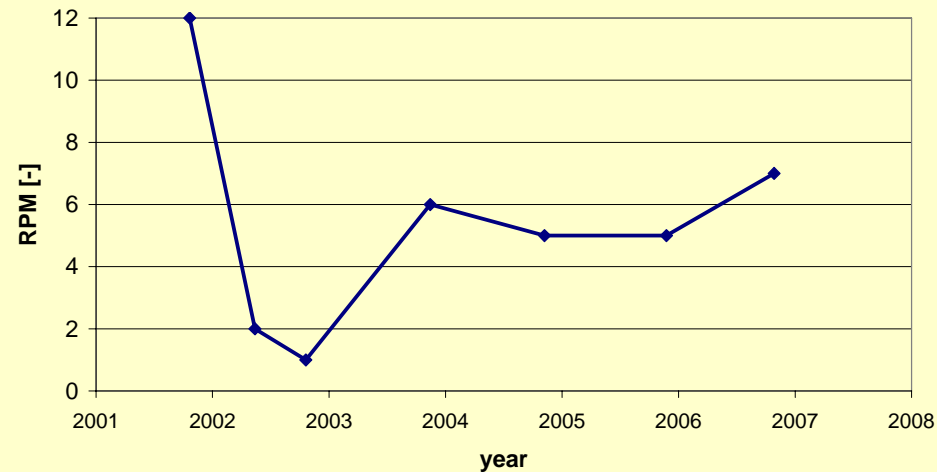


# RPM results individual location

### Turbidity data RPM 100 mm AC

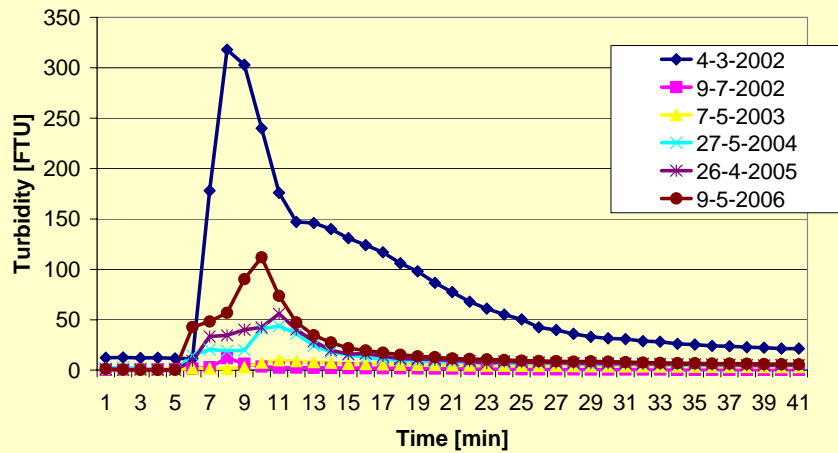


### RPM data 100 mm AC

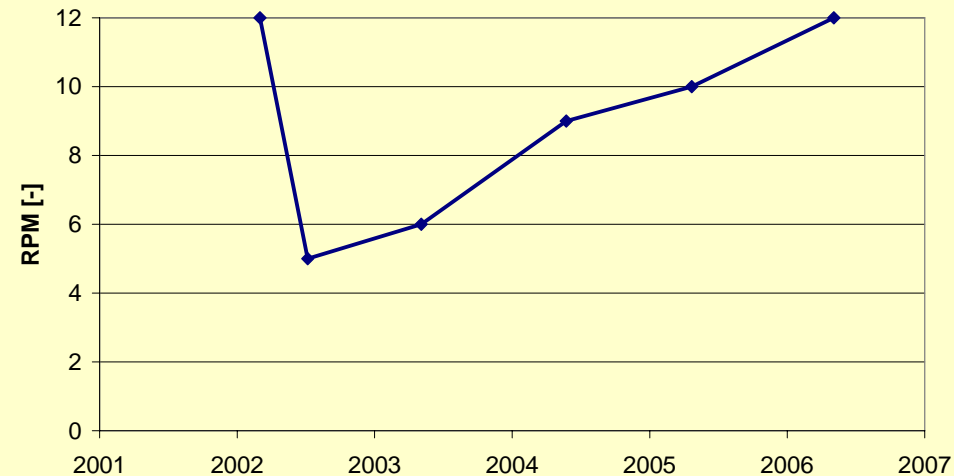


# RPM results CI-location

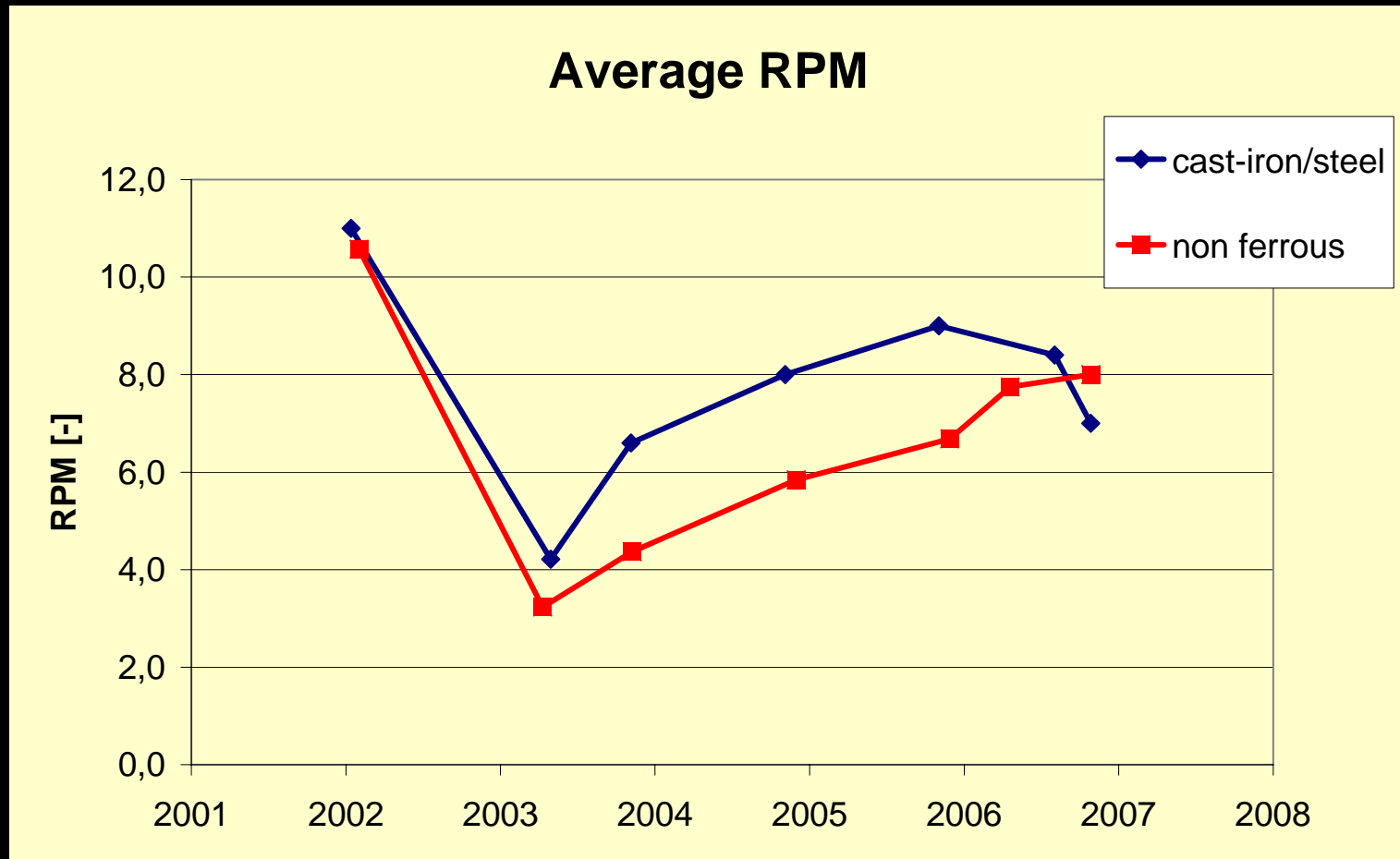
### Turbidity data 4" CI



### RPM data 4" CI

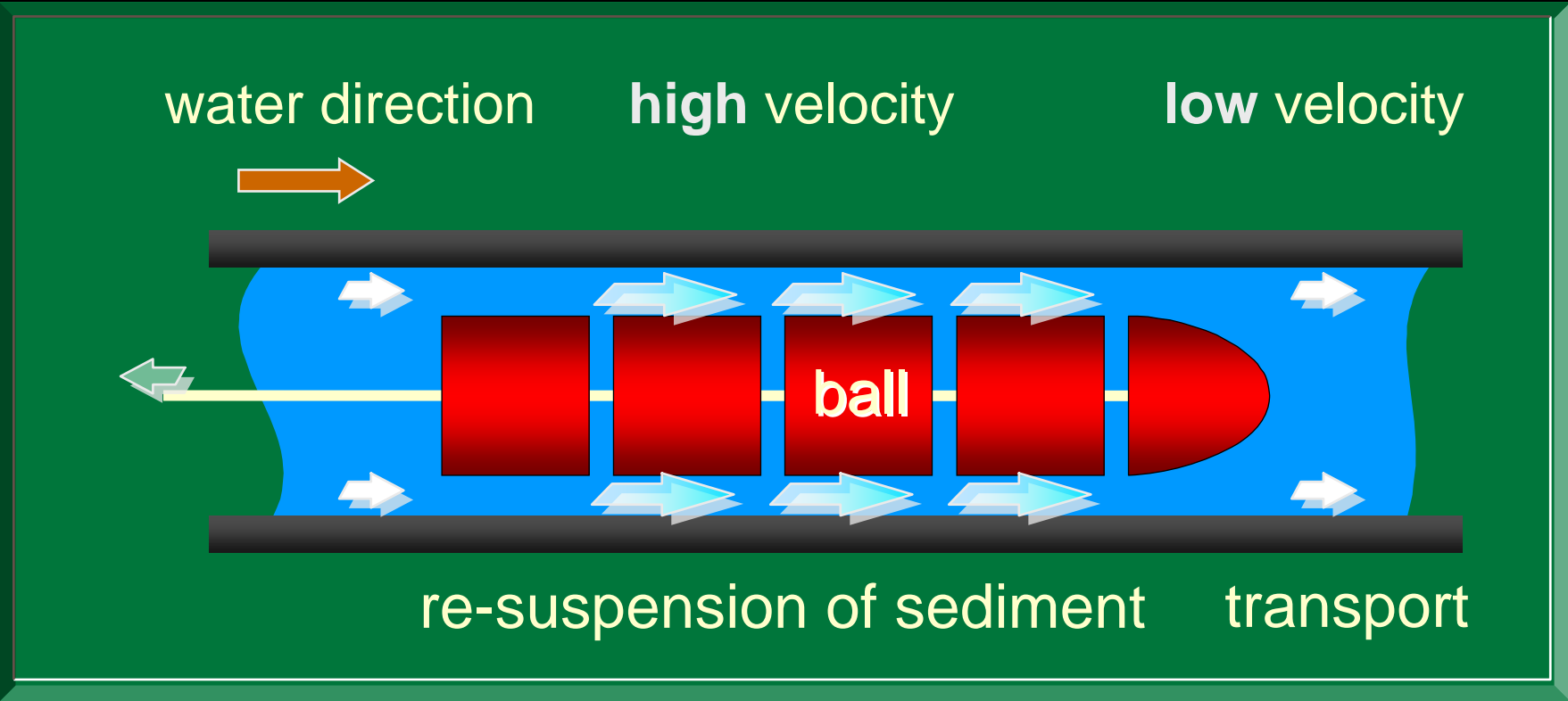


# Overall results





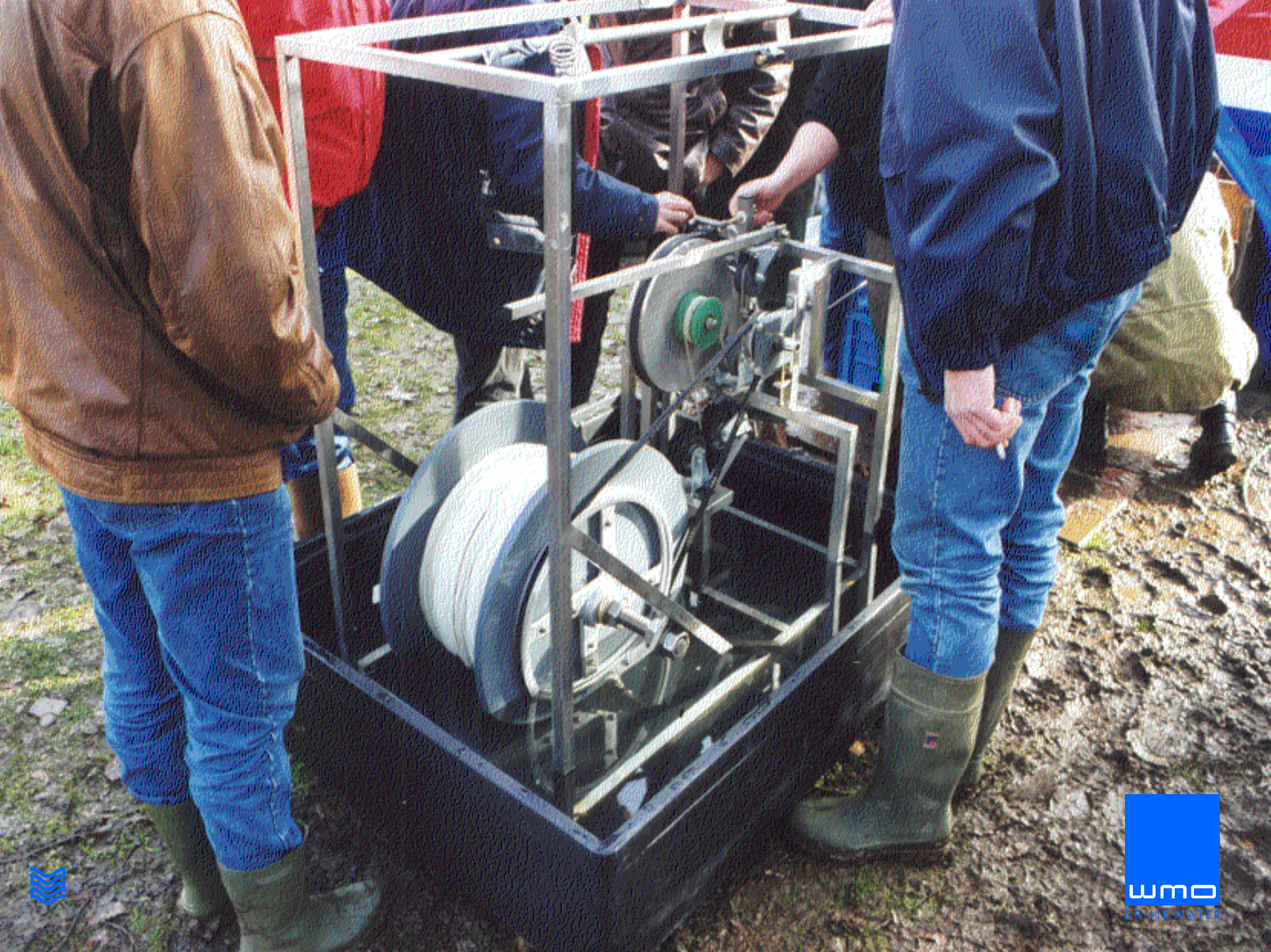
# Innovative cleaning methods















# Water/air scouring

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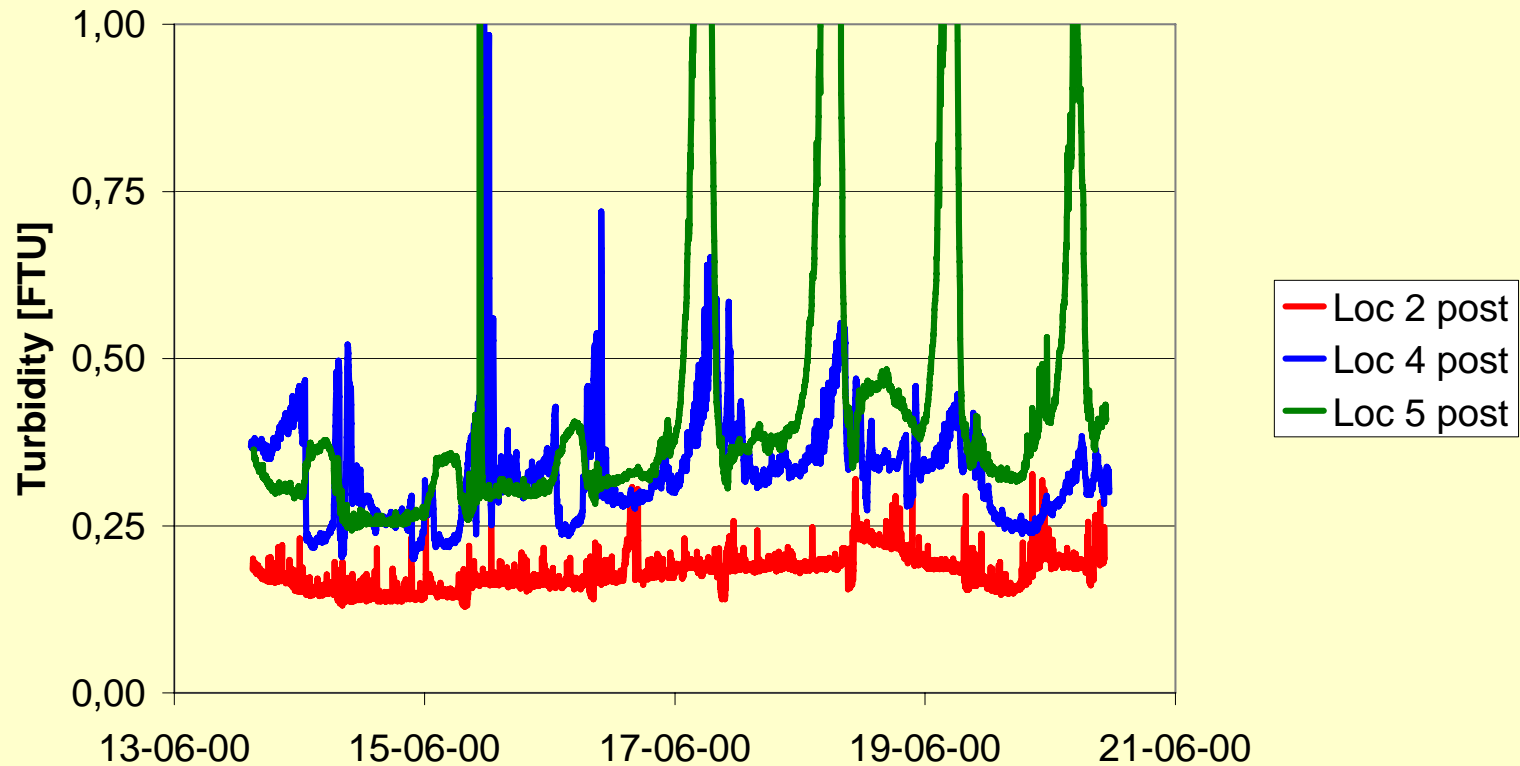


# Pigging



# Result of aggressive cleaning cast iron (pigging or W/A scouring)

## Post-cleaning turbidity



# Third stage: prevent sedimentation

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- Design drinking water networks instead of fire fighting networks
- Main characteristics of drinking water network
  - Looped transport network
  - Branched district areas with up to 200 connections
  - Velocity in branches 0,4 m/s once a day
- Result: 'self cleaning' network
- Large draw back: less fire fighting capacity

# 'Self cleaning' network

Looped network  
 $V = 0,07$  m/sec



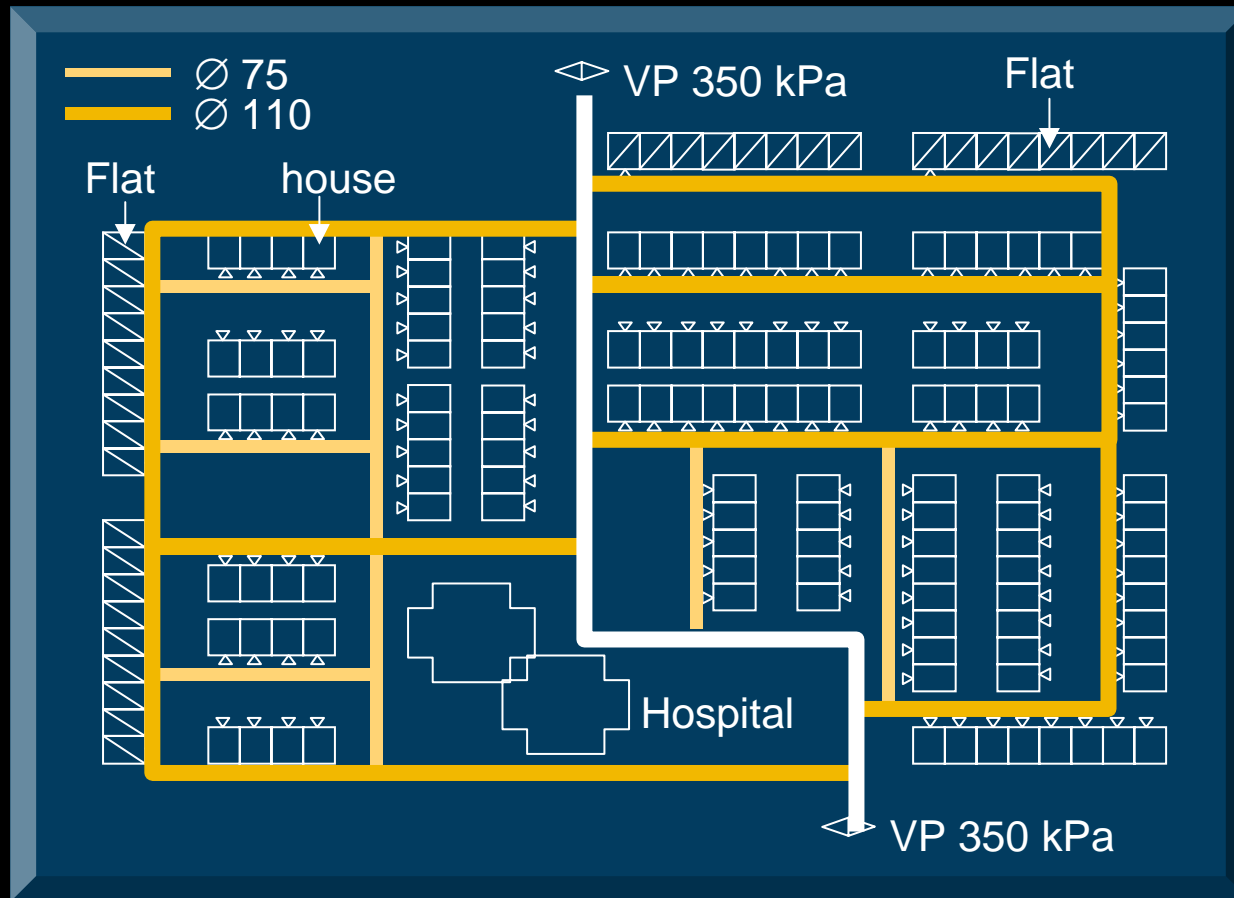
Looped network (disturbed)  
 $V = > 0,4$  m/sec



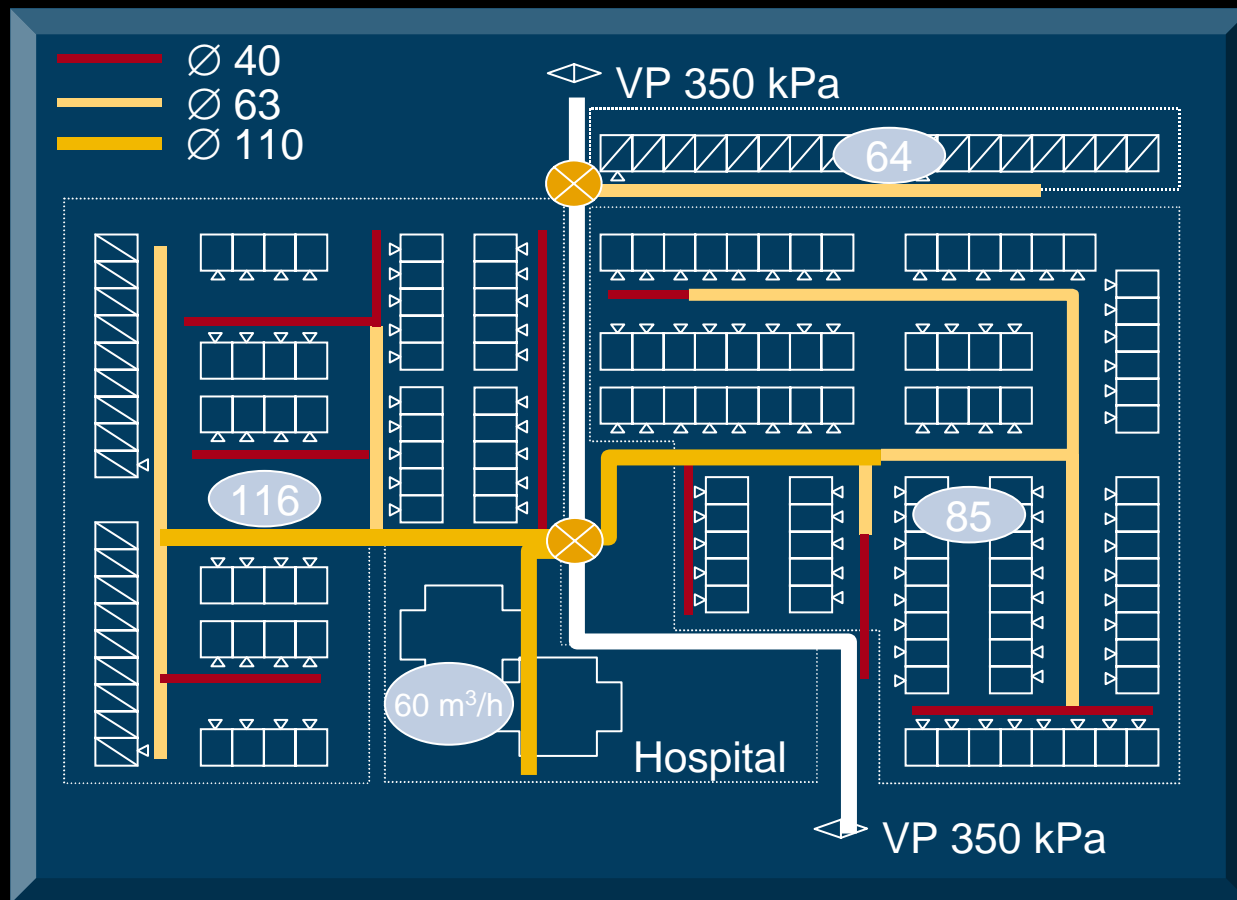
Drinking water network (self cleaning)  
 $V = > 0,4$  m/sec



# Conventional network



# 'New' network



# Pumping stations and water transport

Water quality aspects of drinking water  
networks

ct5550





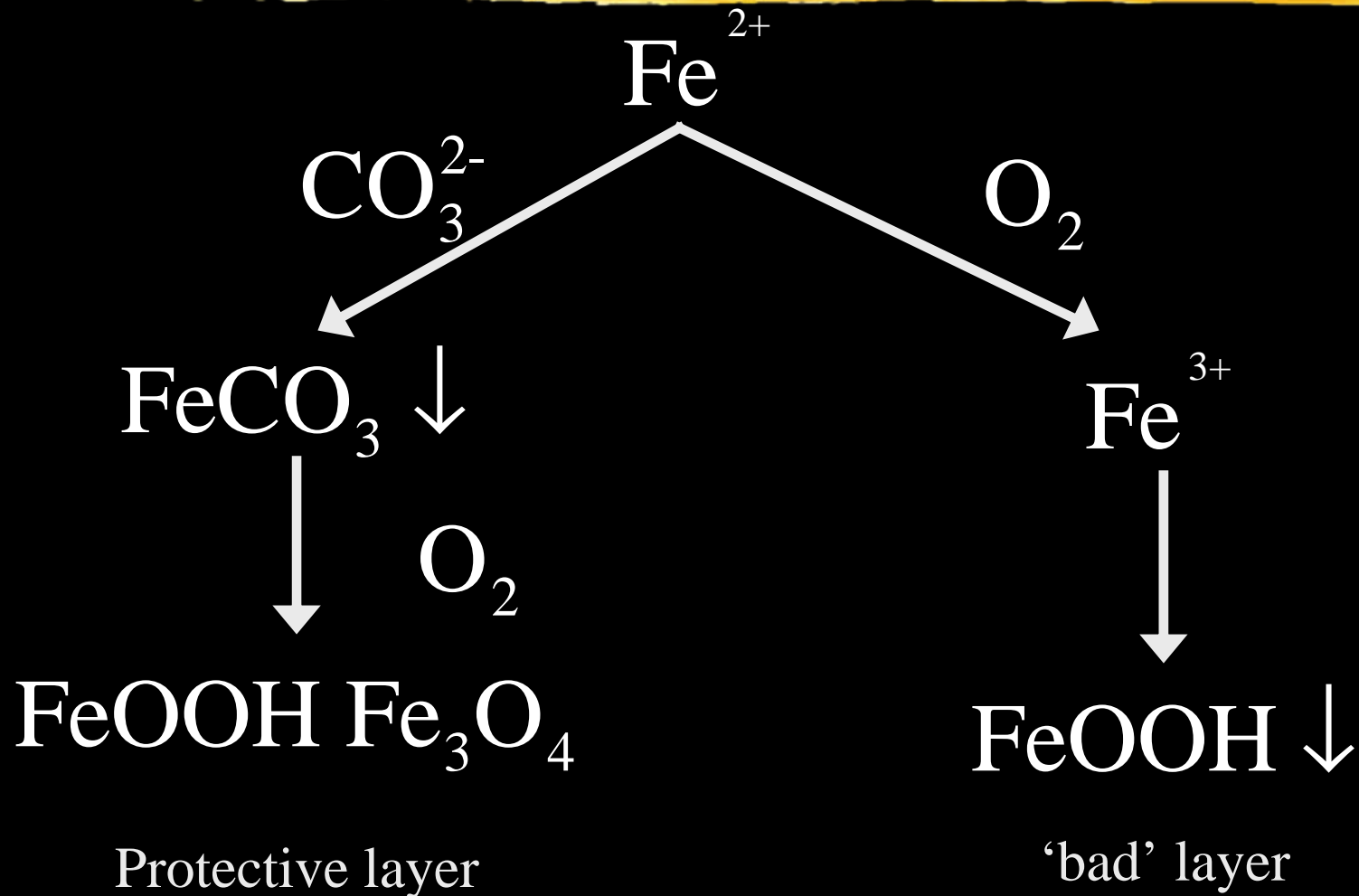


# Chemical processes

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- Materials used in drinking water transport and distribution
  - Cast Iron
  - Asbestos Cement
  - Plastics as PVC and PE
- Materials used in house plumbing
  - Lead
  - Copper
  - Plastics, mainly PE
- Other materials as concrete, (galvanised) steel, GRP act chemically in similar ways

# Cast Iron: corrosion products

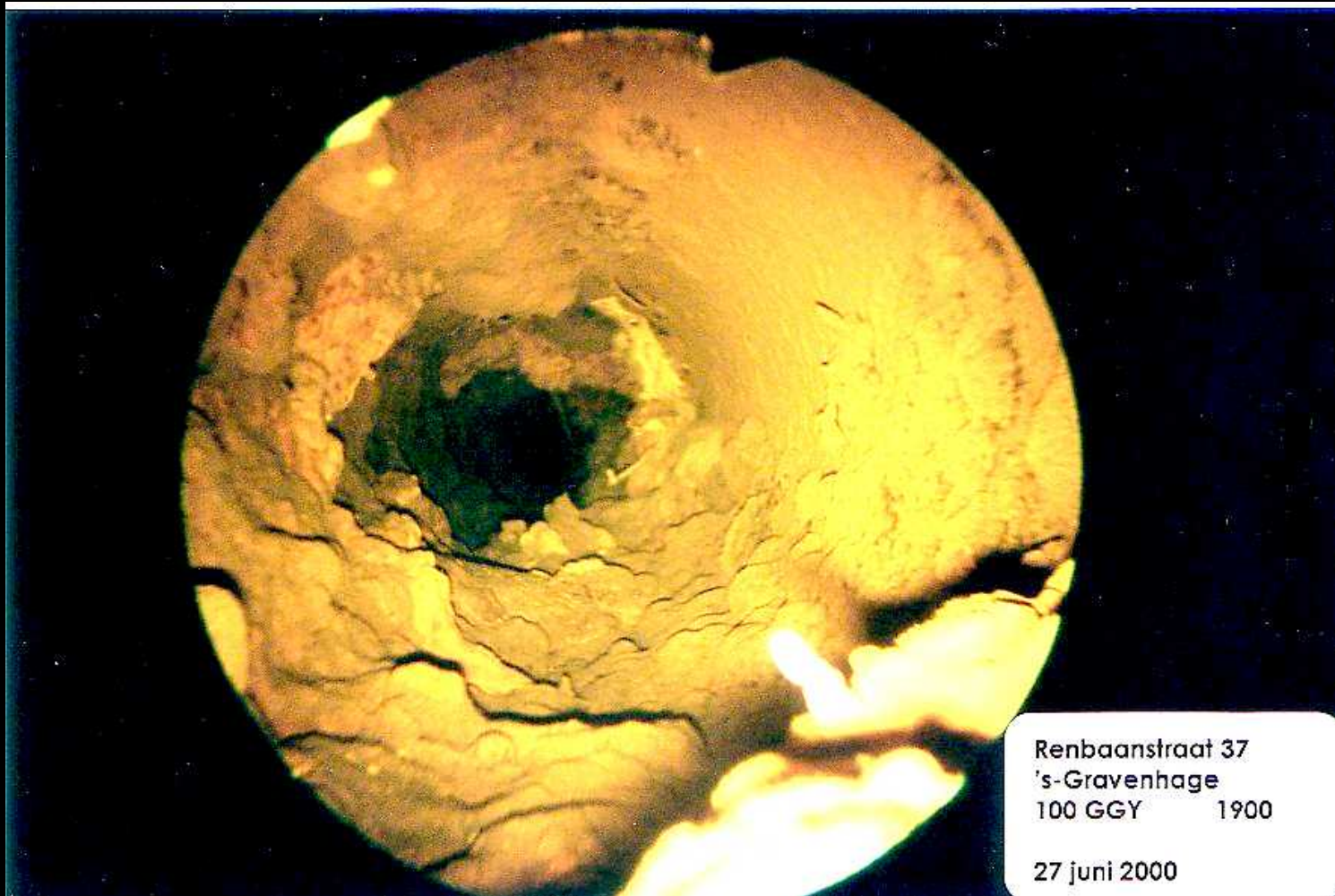


# Cast Iron: corrosion problems

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- Graphitisation: weakening of pipe material and strength
- Loss of hydraulic capacity due to voluminous corrosion products
- Deterioration of water quality
  - Discolouration
  - Biofilm formation

# Ø100 Cast Iron (1900)



# Cast Iron: prevention of corrosion

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- Water composition at pumping station
  - $\text{pH} > 7,5$
  - $\text{TIC} > 2 \text{ mmol/l}$   
(Total Inorganic Carbon  $\cong [\text{HCO}_3^-]$ )
  - $(\text{Cl}^- + 2\text{SO}_4)/\text{TIC} < 1$
- Coatings
  - Cement mortar lining
  - PE-sleeves

# Asbestos Cement

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- Corrosion process: Leaching of Calcium Hydroxide
- Corrosion problems:
  - Loss of strength
  - Increase of pH
  - Release of asbestos fibres
  - Precipitation of calcium carbonate

# Prevention AC corrosion problems

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- Water composition
  - $-0,2 < SI < 0,3$
  - $TIC > 2 \text{ mmol/l}$

$$SI = \log \frac{[Ca^{2+}][CO_3^{2-}]}{[Ca^{2+}]_{eq}[CO_3^{2-}]_{eq}} = pH - pH_{eq}$$

# Plastics

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- No chemical processes known at the moment
- (limited) Lead destabilisation at the start of the life time



# Lead and copper

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- Corrosion process: release of lead
- Corrosion problem: public health
- Prevention
  - Water composition:  $\text{pH} > 7,8$
  - Coating
  - replacement

# Optimal composition of drinking water

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- $7.8 \text{ and } (0,38 \text{ TIC} + 1.5 [\text{SO}_4^{2-}] + 5.3) < pH < 8.3$
- $TIC > 2$
- $-0.2 < SI < 0,3$
- $\frac{[\text{Cl}^-] + 2[\text{SO}_4^{2-}]}{TIC} < 1$

$$TIC = [\text{HCO}_3^-] \text{ (if } 7.8 < pH < 8.3)$$

# Chemical en biological processes

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- The following slides are optional and can be put in an appendix.

# Biological processes

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- Biological water quality will change in the distribution system
- Changes are governed by:
  - initial water quality
  - chemical and biological processes
  - hydraulics
  - pipe material
  - system integrity

# Goal in-pipe water quality control

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- Hygienic water safety control
  - limitation of regrowth
  - prevention of recontamination
- Controlling chemical and physical composition

# Hygienic water safety control: limitation of regrowth

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- Disinfectant residual
  - May have some effect
  - Not applied because of adverse effects (disinfection by-products, taste, odour)
- Controlling the source of regrowth problems
  - biologically unstable water
  - biologically unstable pipe materials

# Causes for degradation of the microbial in-pipe water quality

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- biofilm formation
  - attachment and growth of micro-organisms onto surfaces exposed to drinking water
  - entrapment of (inorganic) particles in biofilm
- sediment accumulation
  - environment for multiplication of micro-organisms (Aeromonads, invertebrates)

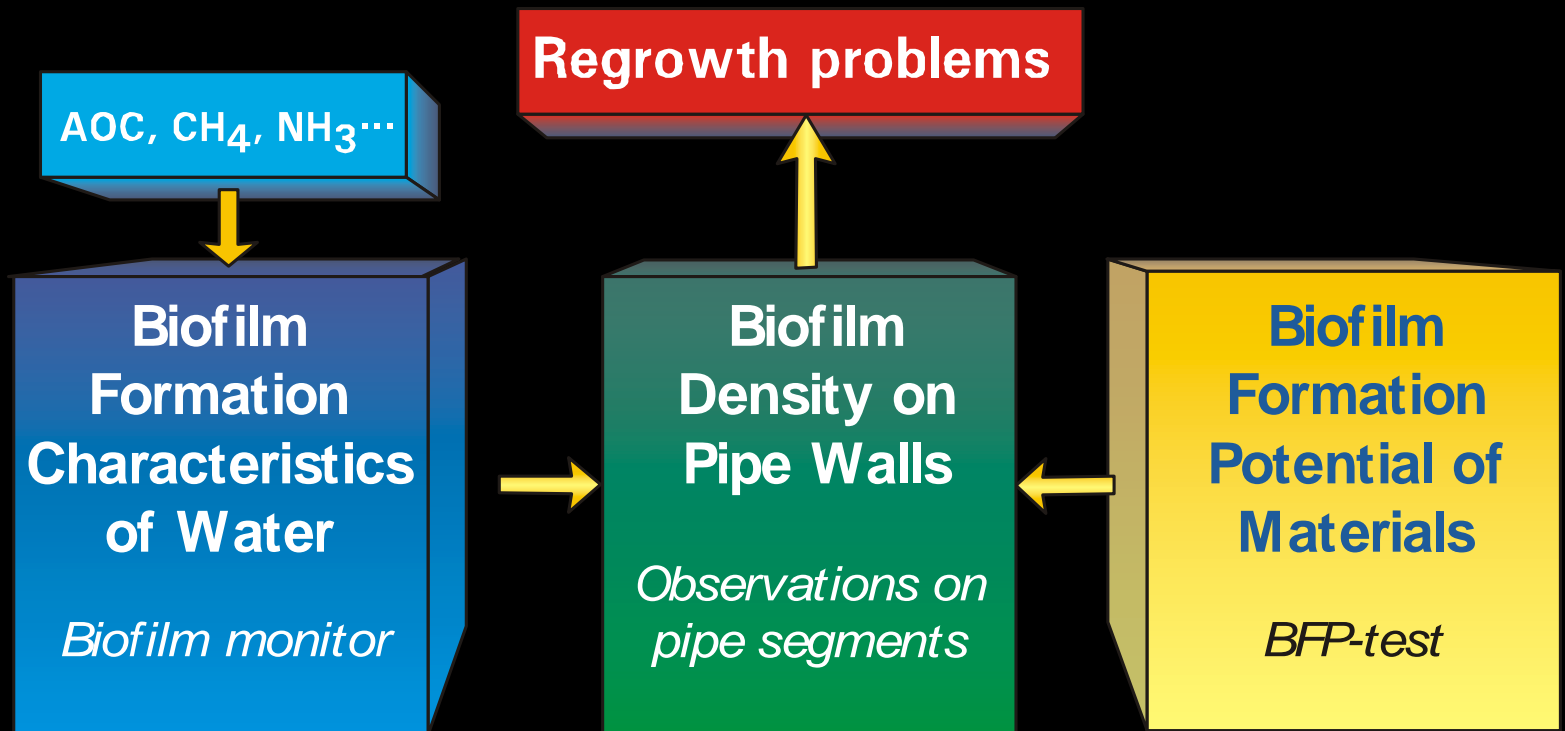
# Biofilm formation

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- Biofilm formation is promoted by microbial utilisation of biodegradable compounds
- Sources for biodegradable compounds:
  - treated water
  - pipe material in contact with drinking water



# The Unified Biofilm Approach

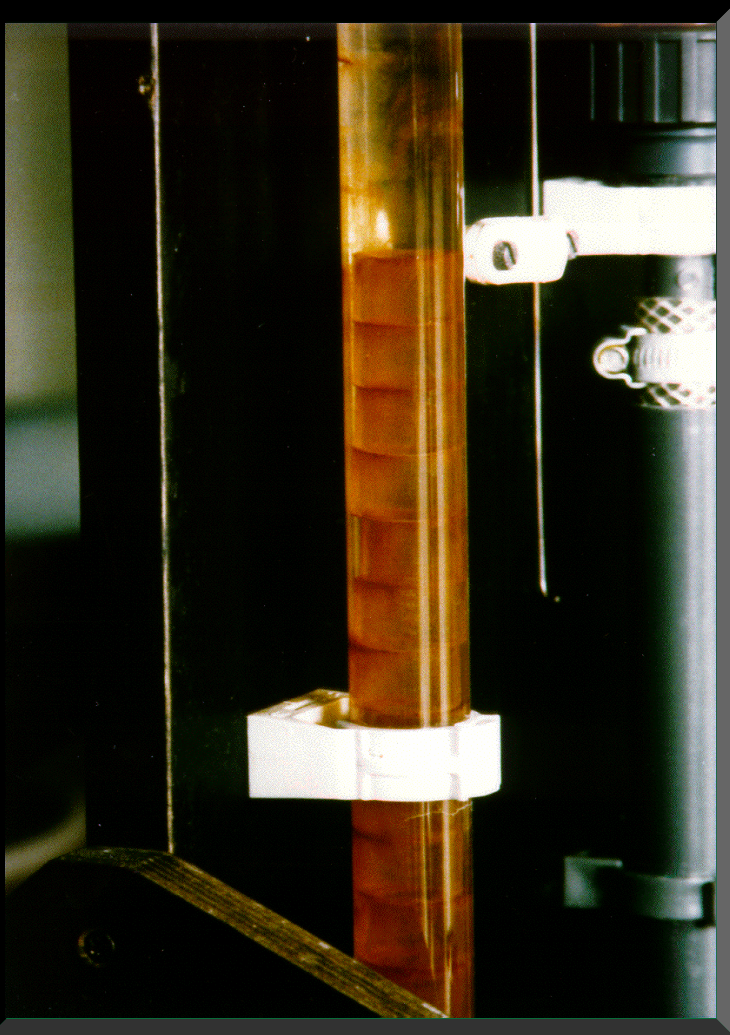


# Assessing biofilm formation

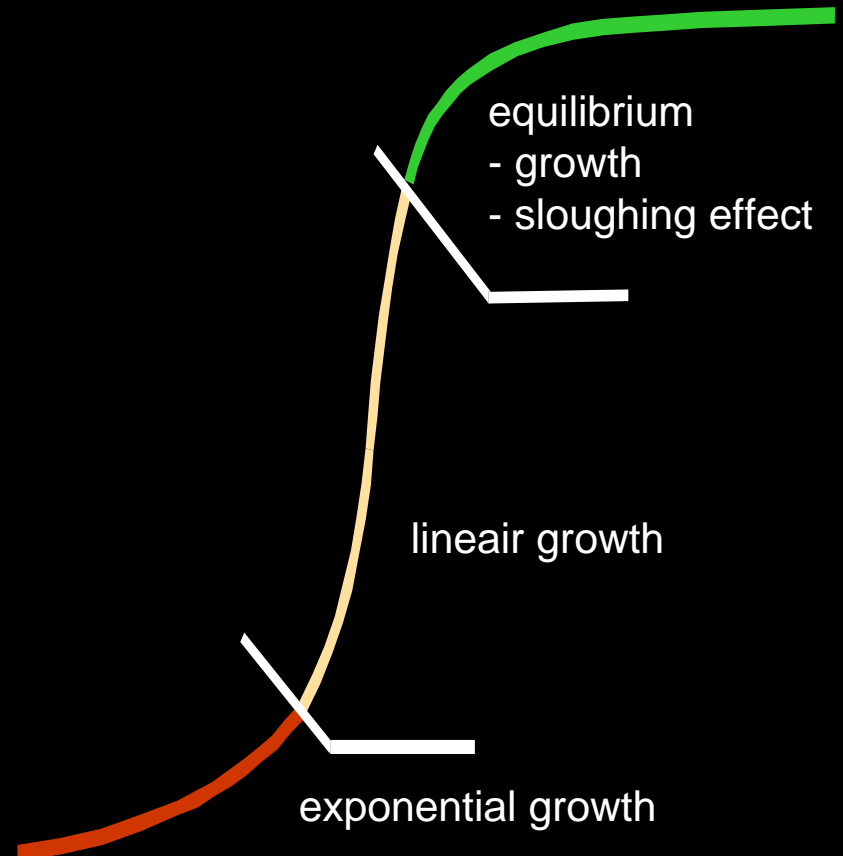
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- Kiwa Biofilm Monitor
  - assessment of rate (BFR) and extent (BFP) of biofilm formation on surfaces exposed to water under test
    - BFR = Biofilm Formation Rate ( $\mu\text{g ATP}/\text{cm}^2\cdot\text{d}$ )
    - BFP = Biofilm Formation Potential ( $\mu\text{g ATP}/\text{cm}^2$ )
  - provides information on the biostability of the water
- Biomass Production Potential (BPP) test
  - assessment of the growth promoting characteristics of materials in slow sand filtrate
  - $\text{BPP} = \text{BFP} + \text{Suspended Biomass Production (SBP)}$

# Biofilm Formation Rate (BFR) in water: Kiwa Biofilm Monitor



## Biofilm formation



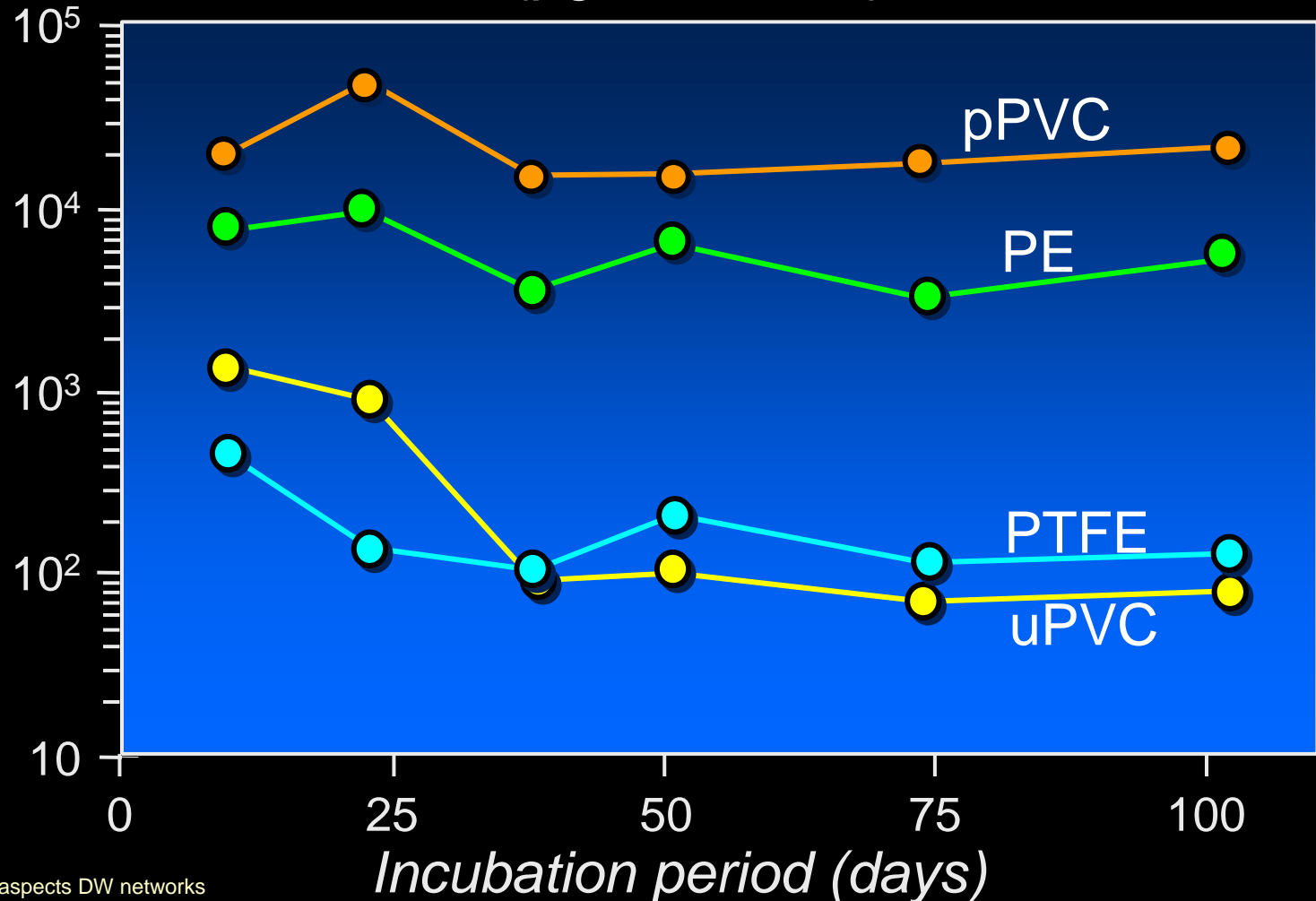
# Biomass Production Potential (BPP)



- BPP of materials in contact with (tap) water
- BPP/BFP test
  - Slow sand filtrate (600 ml)
  - 12 pieces of material
  - $S/V = 0.15 \text{ cm}^2/\text{cm}^3$
  - Duplicate flasks
  - Incubation at 25 °C
  - Period: 16 weeks
  - Biomass parameters:
    - - ATP
    - - HPC
    - - Legionella, ..

# Biofilm Formation on materials in slow sand filtrate (BFP-test)

*Biofilm Concentration (pg ATP / cm<sup>2</sup>)*



# The Unified Biofilm Approach: a framework for evaluation

