CT4471-OCW DRINKING WATER TREATMENT 1 (2006-2007) (4383-2006OCW) > CONTROL PANEL > PREVIEW ASSESSMENT: AERATION VERSION 2



Preview Assessment: Aeration version 2

Aeration version 2 Name

Instructions Answer the questions in small groups (2 persons). Think well before answering and consult your lecture notes or othe

sources if necessary. The test can only be made once.

Multiple Attempts This Test allows multiple attempts.

Force Completion This Test can be saved and resumed later.

Question Completion Status:

Question 1 10 points Save

Which gas transfer system maches with the given photographs?





- A. Spray aeration
- B. Cascade aeration
- C. Tower aeration
- D. Plate aeration













Question 2 10 points Save

Which words belong to which aeration system?

Cascade aeration.

Tower aeration.

A. orifices.

	- ▼ Plate aeration ▼ Spray aeration.	B. packing media. C. droplets. D. weirs.		
Question	3		10 points	Save
	Nett gass transport takes place from water to air	until the gas concentration in water is equa	ll to	
	∩ RQ			
	saturation concentration.			
	C zero			
	gas concentration in air.			
Question	ı		10 points	Save
	The saturation concentration is calculated with the	ne next formula:		
	kD is smaller at a higher temperature			
	13 Smaller at a higher temperature			
	and the same			
	$c_i = k_D \cdot c_i$			
	0.5			
	O True			
	C False			
Question	i		10 points	Save
	Gasses with a low k_D-values hardly dissolve in	water and are therefore difficult to remove	•	
	the water.			
	C True			
	C False			
Question			10 points	Save
	With an increasing water temperature the satural	tion concentration		
	C Decreases			
	C increases			
	remains the same			
Question			10 points	Save
	Which equation need to be used if the variation on neglected?	of the gas concentration in air cannot be		
	equilibrium equation			
	kinetic equation			

mass balance

Question 8

10 points

Save

Fill in the blanks:

k_2 is the gas transfer coefficient. The larger the contact surface area between air and water and the renewal of this surface area the the gas transfer and the the gas transfer coefficient.

- worse, higher
- worse, smaller
- better, higher
- better, smaller

Question 9

10 points

Save

K is the efficiency of a gas transfer system. Which formula belongs to which basic system?

- 🔻

$$K_1 = 1 - \exp(-k_2 \cdot t)$$

- ▼

$$K_2 = \frac{1}{1 + \frac{1}{k_2 \cdot t}}$$

- 🔻

$$K_{3} = \frac{1 - \exp\left(-k_{2} \cdot t \cdot \left(1 + \frac{k_{D}}{RQ}\right)\right)}{1 + \frac{k_{D}}{RQ}}$$

$$K_4 = \frac{1 - \exp\left(-k_2 \cdot t \cdot \left(1 - \frac{k_D}{RQ}\right)\right)}{1 - \frac{k_D}{RQ} \cdot \exp\left(-k_2 \cdot t \cdot \left(1 - \frac{k_D}{RQ}\right)\right)}$$

- 🔻

$$K_5 = \frac{1}{1 + \frac{1}{k_2 \cdot t} + \frac{k_D}{RQ}}$$

- A. Plug flow, co-current flow and variable gas concentration in air.
- B. Complete mixed system with variable gas concentration in air.
- C. Plug flow with constant gas concentration in air.
- D. Plug flow, counter current flow and variable gas concentration in air.
- E. Complete mixed system with constant gas concentration in air.

Question 10 10 points Save

The removal of carbondioxide by one cascade is independent on fall height.

- True
- False

Question 11 10 points Save

The composition of air is given in table. For the removal of methane from groundwater a water company uses cascade aeration. The aeration consists of 5 stages and the total falling height is 2 m. The concentration of methane in the raw water is 0,8 mg/l and after the first stage 0,54 mg/l.

Table ¿ composition of air (10°C, 101325 Pa).

Gas	Volume percentage [%]
N_2	78,084
O_2	20,948
Ar	0,934
CO ₂ CH₄	0,034
CH ₄	0,00001

Calculate the equilibrium concentration of methane in water at a pressure of 101325 Pa and a temperature of 10°C.

2.96*10-4	ma/l

8.38*10-5 mg/l

2.96*10-6 mg/l

2.34*10-6 mg/l

Question 12 10 points Save

For the removal of methane from groundwater a water company uses cascade aeration. The aeration consists of 5 stages and the total falling height is 2 m. The concentration of methane in the raw water is 0,8 mg/l and after the first stage 0,54 mg/l.

Calculate the methane removal after 5 cascade stages.

0.86

0.33

0.39

0.96

Question 13 10 points Save

Assuming K=0.33 and Kd is 0.034, calculate the value of k_2^*t for only one cascade stage with the assumption that the RQ of a cascade stage is 0.4.

0.49

O.40

0.41

0.52

Question 14 10 points Save

In groundwater treatment aeration and gas transfer is needed to remove methane, carbon dioxide and hydrogen sulfide. Why need these gasses to be removed? Match the right explanation with the gasses.

Methane

	- 🔻	A. to avoid excessive dosing of chemicals during softening.		
	- Carbon dioxide	B. To prevent biological growth in filters.		
	- W Hydrogen sulfide	C. For taste and odour.		
Question 15		(450/ 1)	10 points	Sav
	A cascade stage with a height of 30 cm has a gare necessary to remove at least 60%?	as removal efficiency of 15%. How many st	eps	
	O 4			
	○ 6			
Question 16			10 points	Sav
	What is approximately the RQ of a cascade?			
	○ 0.4			
	O 11			
	O 90			
Question 17	For which gasses is the cascade suitable? Mor	o answers can be right	10 points	Sav
	roi willer gasses is the cascade sultable? Moi	e answers can be right.		
	Removal of chloroform.			
	Addition of oxygen.			
	Removal of carbon dioxide.			
	Removal of methane.			
Question 18	The application of a tower aerator can lead to p	recipitation of calcium carbonate	10 points	Sav
	The application of a terror defact our load to p	roophalion of calcium carsonate.		
	○ True			
	○ False			
Question 19			40 mainta	0
Question 19	The retention time in a tower aerator is practica	ally independent of the water flow.	10 points	Sav
	·			
	○ True			
	○ False			
Question 20			10 points	Sov
นุนธ อนปก 20	Is it necessary to back flush a tower aerator?		το μοιπιε	Sav
	Yes, this is necessary.			
	No, this is not necessary.			

 $\hfill \square$ This is only necessary if iron is present in groundwater.

Save Submit