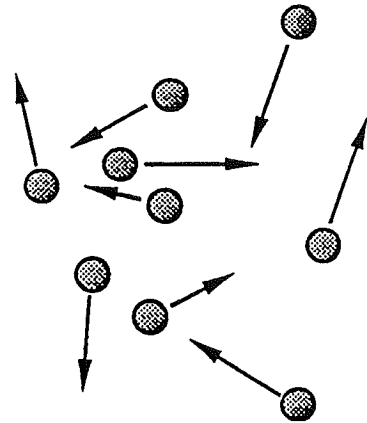
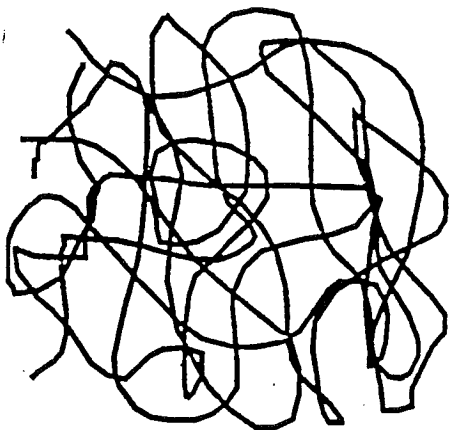


WHAT MAKES A FLUID A BINGHAM OR POWER-LAW FLUID?

BASIC UNIT OF NEWTONIAN FLUID
IS MOLECULE, USUALLY < 100
ANGSTROMS LONG (10^{-6} cm)

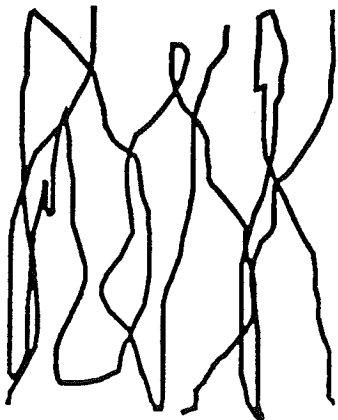


BASIC ORIGIN OF VISCOSITY IN
NEWTONIAN FLUIDS IS
MOLECULAR COLLISIONS, i.e.
EXCHANGE OF MOMENTUM
BETWEEN MOLECULES



BASIC UNIT OF POWER-LAW
FLUIDS IS USUALLY
EXTREMELY LONG POLYMER
MOLECULES - 10^6 ANGSTROMS
OR LONGER
THESE MOLECULES GET TANGLED
- INCREASES APPARENT
VISCOSITY

THEY USUALLY STRETCH OUT
AND UNTANGLE AND HIGHER
SHEAR RATES \rightarrow LOWER
EFFECTIVE VISCOSITY (SHEAR-
THINNING BEHAVIOR)

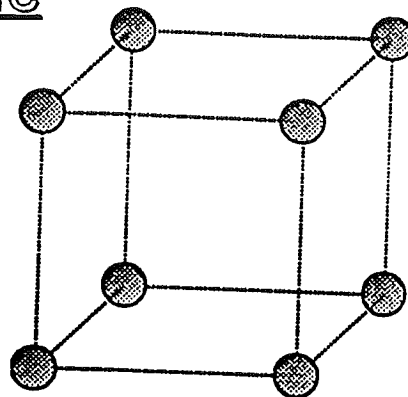


IF THEY CAN'T UNTANGLE AT HIGH
SHEAR RATES \rightarrow HIGHER
EFFECTIVE VISCOSITY (SHEAR-
THICKENING BEHAVIOR)

BASIC UNIT OF BINGHAM PLASTIC
IS USUALLY A CRYSTAL-LIKE
ARRANGEMENT OF SMALL
PARTICLES

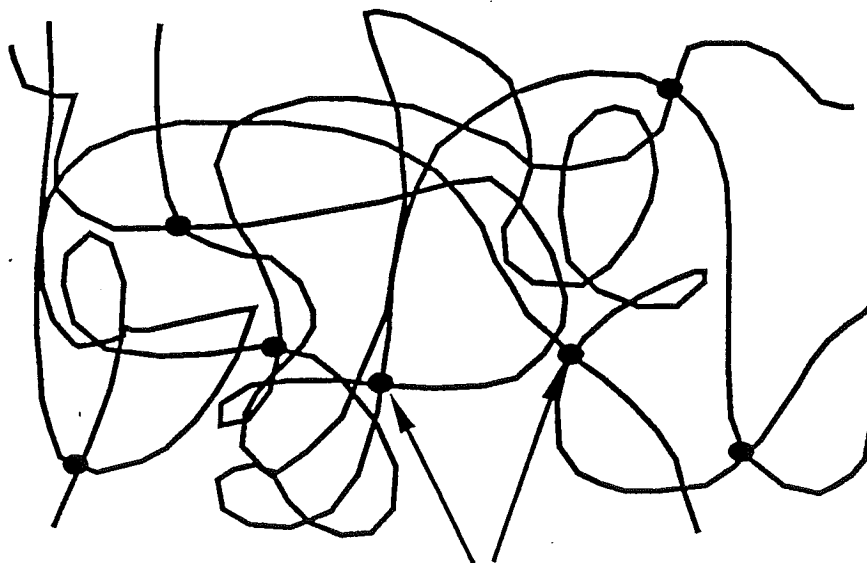
"CRYSTAL" STRUCTURE RESISTS
DEFORMATION AT LOW τ -
BEHAVES LIKE SOLID
HIGHER τ TEARS UP STRUCTURE -
FLUID BEGINS TO FLOW

Alternative structure: polymer gel (below)



BOTH BINGHAM AND POWER-LAW FLUIDS HAVE
MOLECULES OR STRUCTURES THAT CAN BE
BIGGER THAN PORE THROATS IN ROCK
ON PORE SCALE THESE FLUIDS ARE NOT
HOMOGENEOUS

MODELS ARE NOT VERY ACCURATE IN THESE CASES
- ESPECIALLY BINGHAM PLASTIC



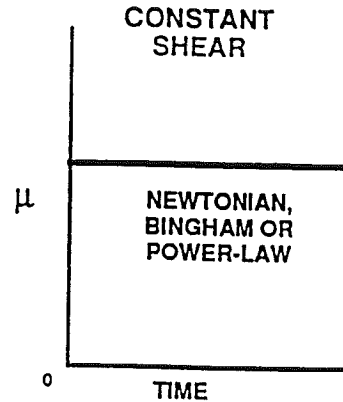
WEAK CHEMICAL OR
ELECTROSTATIC BONDS

STRUCTURE OF
A POLYMER GEL

II. D. 6.

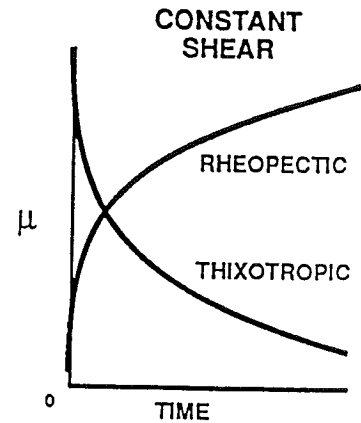
MORE-COMPLEX FLUIDS

FOR NEWTONIAN, BINGHAM AND POWER-LAW FLUIDS, RELATION BETWEEN τ AND $\frac{dV}{dx}$ DEPENDS ON SHEAR RATE, BUT NOT ON TIME



TIME DEPENDENT FLUIDS:

- THIXOTROPIC - VISCOSITY DECREASES WITH TIME AT CONSTANT τ
- RHEOPECTIC - VISCOSITY INCREASES WITH TIME AT CONSTANT τ



VISCOELASTIC - VISCOSITY DEPENDS ON RECENT SHEAR HISTORY