

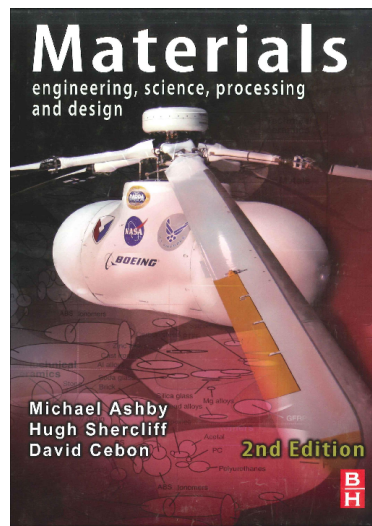
Introduction

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Department of Materials Science and Engineering (MSE)
Delft University of Technology



Book




NASA tests of helicopter rotor blades made with piezo-electric materials



Contents

Introduction, page 3

- Book:
[Materials engineering, science, processing and design](#)
Michael Ashby, Hugh Shercliff, David Sherbon
Elsevier Butterworth-Heinemann (Oxford 2010)
 - Chapters 1-11 
- Supporting lecture notes:
[Based on PowerPoint notes by Prof. John Barnard, University of Pittsburgh](#) (Blackboard)
- Supporting software:
[CES EduPack Eco Edition 2011](#) (Blackboard)



Schedule 2011-2012

Introduction, page 4

- Wednesdays + Fridays, 10:45 am - 12:30 pm
- First class on Friday 9 September
- Last class on Wednesday 26 October
- 14 classes
- Perhaps extra class on Friday 28 October

- Office hours: Fridays 4pm-5pm, appointment by e-mail only (B.J.Thijssse@tudelft.nl), max 15 minutes
- I handle course content and personal requests, not organizational matters



Schedule 2011-2012

Introduction, page 5

#	Day	Date 2011	Book chapter	Special
1	Fri	9 Sep	Introduction, Ch 1	
2	Wed	14 Sep	Ch 2	
3	Fri	16 Sep	Ch 3	
4	Wed	21 Sep	Ch 4	
5	Fri	23 Sep	Ch 4, Ch 5	
6	Wed	28 Sep	Ch 5	
7	Fri	30 Sep	Ch 6	
8	Wed	5 Oct	Ch 6, Ch 7	
9	Fri	7 Oct	Ch 7	
10	Wed	12 Oct	Ch 8	
11	Fri	14 Oct	Ch 9	
12	Wed	19 Oct	Ch 10	
13	Fri	21 Oct	Ch 11	
14	Wed	26 Oct	Ch 11	
15	Thu	28 Oct		reserve
		XX Nov	Examination	
		XX Jan	Re-examination (2nd opportunity)	



Homework, Testing, Exam

Introduction, page 6

- Homework
 - Yet to be decided
- Testing
 - In class exercises for self-assessment
 - Peer Instruction (Harvard, USA)
 - Laptops, iPads, smart phones for internet connection
- Exam
 - Open Book, **no communication devices allowed**
(iPads, smartphones)



Lab class

Introduction, page 7

- Lab class Materiaalkunde I (WB6101-11 D2)
 - Self-contained and independent part of the course
 - Final grade for WB6101-11:
 - 80% Exam grade + 20% Lab grade
 - Only if both are 5 or higher
 - Lab class organized by Dr Ton Riemslag and team



Rules

Introduction, page 8

- Use of laptops, iPads, e-readers in class: allowed – but no sound
- Use of phones in class: allowed only for connection to the Internet – but no sound



What's so important about materials?

Introduction, page 9

- **Materials are relevant to your life**
 - Think about it
 - From baby carriage to rollator



Why should engineers bother?

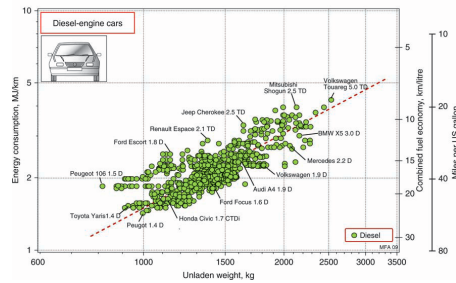
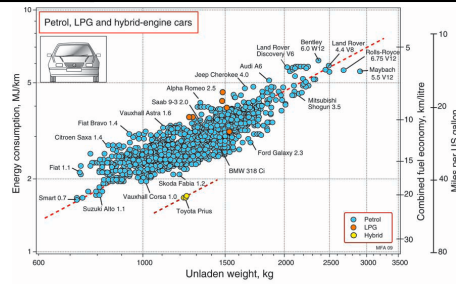
Introduction, page 10

- **Engineers make things**
- **They make them out of materials**
 - What are important materials properties in mechanical engineering?
 - Why are different materials different?
 - How will materials behave under extreme conditions?
 - How can engineers change or create materials?
 - How can the best materials for a given device be selected?



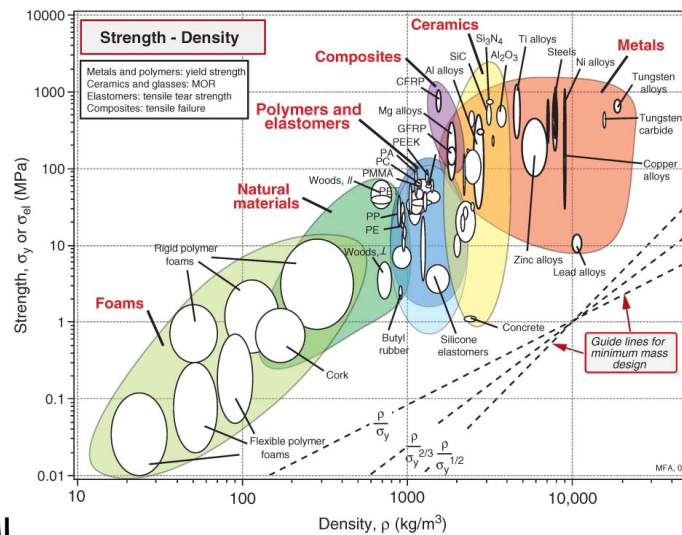
Materials Properties and Choices

Introduction, page 11



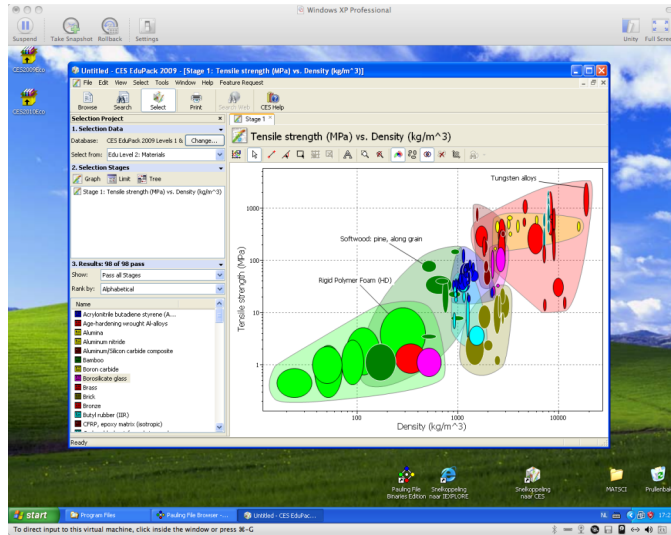
Materials Properties and Choices

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CES Edupack

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Materials in the news

Introduction, page 14

Ferrari roept 1250 auto's terug

ANP/AFP op 02 september '10, 08:14, bijgewerkt 02 september '10, 09:52



Felipe Massa poseert bij een Ferrari 458 Italia (EPA)

ROME - De Italiaanse autofabrikant Ferrari roept wereldwijd 1250 auto's van het type 458 Italia terug naar de garage, omdat de voertuigen in brand kunnen vliegen. Dat heeft Ferrari bekendgemaakt.

De terugroepactie gaat over auto's die zijn gemaakt voor juli 2010. Ferrari begon vorige week een onderzoek, nadat vier voertuigen van het type 458 Italia in brand waren gevlogen. Dat gebeurde in de Verenigde Staten, China, Frankrijk en Zwitserland.

De problemen worden veroorzaakt door een lijn waarmee de hiteschilden zijn aangebracht. Die lijn kan vlam vatten bij extreme hitte.



<http://www.mrs.org>

MRS Bulletin August 2011

Mussel adhesive inspires tough coating for living cells



Taking a cue from shellfish, scientists in Korea have developed a coating that could protect living cells during long exposures to intense heat, dryness and other hostile conditions. In the *Journal of the American Chemical Society*, Insung S. Choi and colleagues report that an organic material called polydopamine, which is a chemical analogue of mussel adhesive, protected yeast cells from cell-digesting chemicals while slowing down their rate of cell division, thereby prolonging their life cycles. The researchers liken the coating to the armor that encloses the spores of some bacterial cells, making them difficult to kill. Polydopamine could be used to encapsulate individual cells to create tiny chemical probes, single-cell chemical factories, and perhaps armor for transplanted cells used in anti-cancer therapies.



A quasicrystalline-coated pan (right) compared with a teflon-coated pan (left).

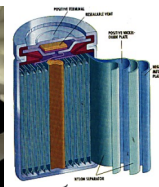
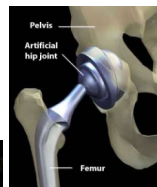


Materials in a global perspective

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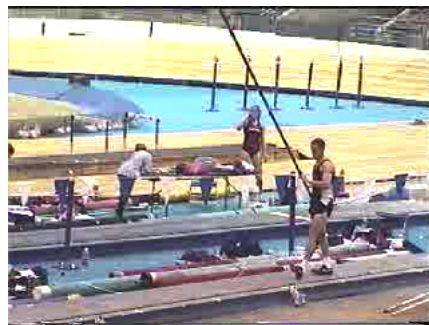


- Communication
- Transportation
- Information
- Energy
- Art and design
- Consumer products
- Sports
- Medicine
- Security
- Infrastructure

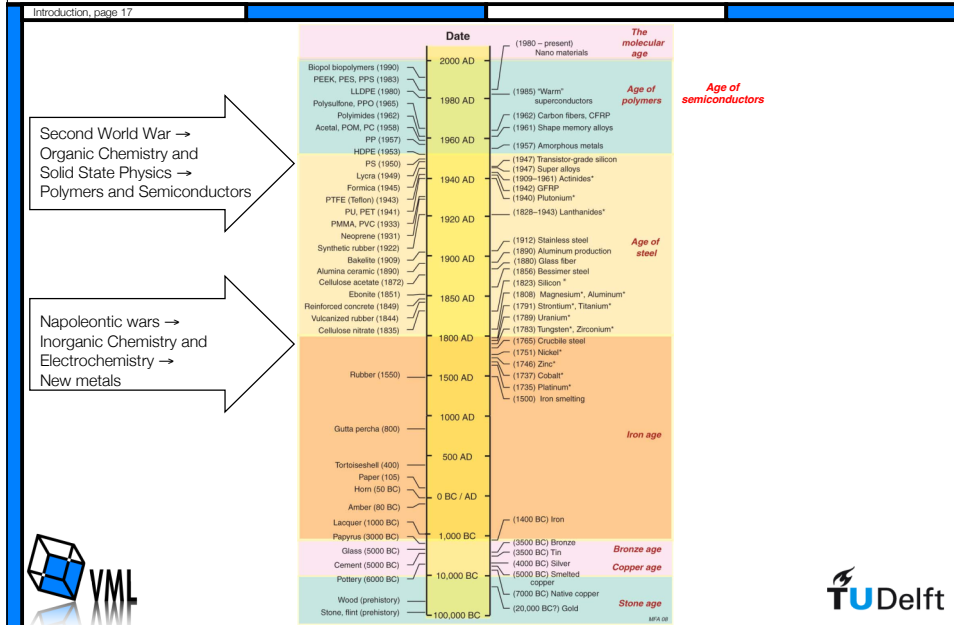


Man depends on materials

Introduction, page 16



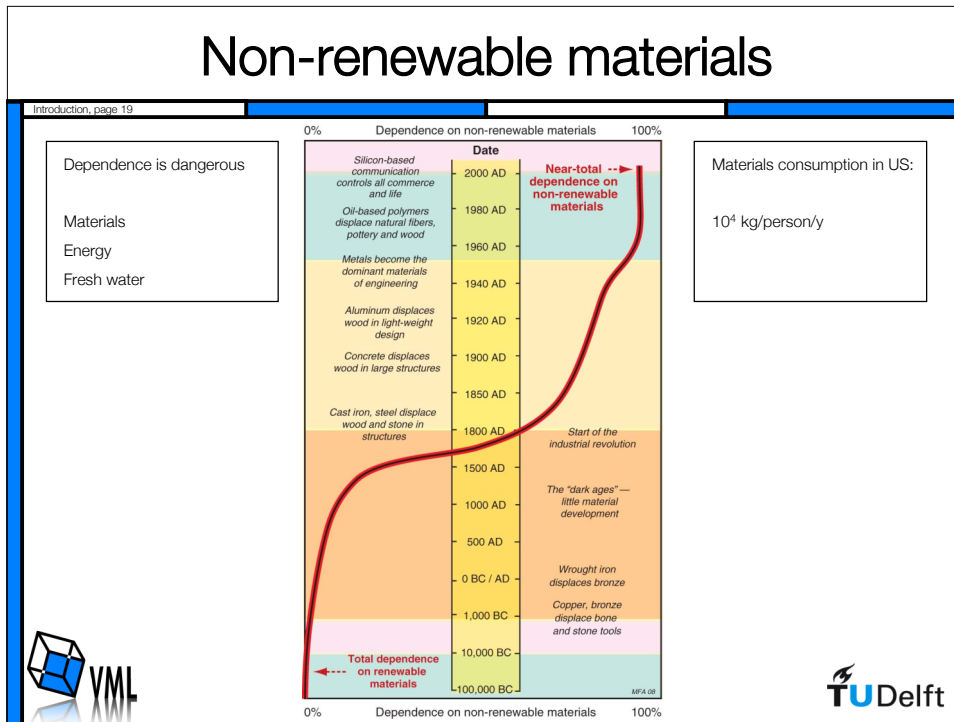
Mankind depends on materials



Critical elements



Non-renewable materials



Class testing

Introduction, page 20

- Example of Peer Instruction
- Suppose that I have just presented this power consumption table

Primary power consumption	
	kW
Electric razor	0.03
Person	0.1
Candle	0.1
Laptop	0.3
Bicycle (extra food)	0.4
Kitchen gas burner	2
Vacuum cleaner	3
Full train per seat	8
Hot shower	10
Medium CV heater	20
Full car per seat	20
Full plane per seat	350

A

TP Anywhere: Window → Showbar

TP Anywhere: Window → Responseware
(B.J.Thijssse@tudelft.nl, Gavia incident)

Answering via the internet

Introduction, page 21

<http://www.nvypoll.com>
User: B.J.Thijssse@tudelft.nl
Password: Gavia incident

Teacher login

ResponseWare
Session ID: 467452 Get New Session ID
Welcome: Logout
Status
Number of Connections: 1
Average Response Time: 523 ms

Student login

Welcome
Enter Session ID
Join
Manage Account
For a better user experience try our ResponseWare mobile app.

ResponseWare Web
About ResponseWare
Contact Us
Terms of Use
Secure Login Assistance
www.funtestinc.com

Welcome
Device ID
Device ID
First Name
Last Name
User Data
Continue
ResponseWare Web
About ResponseWare
Contact Us
Terms of Use
Secure Login Assistance
www.funtestinc.com

Welcome to Session 467452
Please wait for your presenter to open polling.
Feedback
Device ID
Device ID
Session ID = 467452
www.funtestinc.com

Which circuit produces more light?
1 A
2 B
3 Both the same amount
Answer: Test View
Feedback
Device ID
Device ID
Session ID = 467452
www.funtestinc.com

Which circuit produces more light?
Answer
A 33.33%
B 66.67%
Both the same amount 0%
Feedback
Device ID
Device ID
Session ID = 467452
www.funtestinc.com



What is the power consumption in Holland, expressed as a 24/7 continuous rate?

Introduction, page 22

1. 60 W per person
2. 600 W per person
3. 6 kW per person
4. 60 kW per person



Question List --> Open Question List
Window --> Showbar



Materials in a high-level road bike

Introduction, page 23



Wilier Cento (2006)



Which material is *not* present in the bike?

Introduction, page 24



1. Metal
2. Semiconductor
3. Gas
4. Composite
5. Liquid
6. Ceramic
7. Liquid crystal
8. Polymer
9. They're all there



Useful background information

Introduction, page 25

- Periodic system
- Units and conversion
- Energy data
 - From: Jo Hermans, *Energie survival gids* (2009)



Periodic system

Introduction, page 26

Periodic Table of the Elements

Legend: Solids (white), Liquids (blue), Gases (pink), Artificially Prepared (yellow)

Example element Fe: Atomic Number 26, Symbol Fe, Name Iron, Atomic Weight 55.845

PERIOD	GROUP IA	IIA	IIIA	IVA	VA	VIA	VIIA	VIIIA	IB	IIB	IIIB	IVB	VB	VIB	VIIA	VIII					
1	H Hydrogen (1.008)															He Helium (4.003)					
2	Li Lithium (6.941)	Be Beryllium (9.012)														B Boron (10.811)	C Carbon (12.011)	N Nitrogen (14.007)	O Oxygen (15.999)	F Fluorine (18.998)	Ne Neon (20.180)
3	Na Sodium (22.990)	Mg Magnesium (24.305)														Al Aluminum (26.982)	Si Silicon (28.086)	P Phosphorus (30.974)	S Sulfur (32.06)	Cl Chlorine (35.453)	Ar Argon (39.948)
4	K Potassium (39.098)	Ca Calcium (40.078)	Sc Scandium (44.956)	Ti Titanium (47.867)	V Vanadium (50.942)	Cr Chromium (51.996)	Mn Manganese (54.938)	Fe Iron (55.845)	Co Cobalt (58.933)	Ni Nickel (58.693)	Cu Copper (63.546)	Zn Zinc (65.38)	Ga Gallium (69.723)	Ge Germanium (72.63)	As Arsenic (74.922)	Se Selenium (78.972)	Br Bromine (79.904)	Kr Krypton (83.80)			
5	Rb Rubidium (85.468)	Sr Strontium (87.62)	Y Yttrium (88.906)	Zr Zirconium (91.224)	Nb Niobium (92.906)	Mo Molybdenum (95.94)	Tc Technetium (98)	Ru Ruthenium (101.07)	Rh Rhodium (102.905)	Pd Palladium (106.42)	Ag Silver (107.868)	Cd Cadmium (112.411)	In Indium (114.818)	Sn Tin (118.71)	Sb Antimony (121.757)	Te Tellurium (127.6)	I Iodine (126.905)	Xe Xenon (131.29)			
6	Cs Cesium (132.905)	Ba Barium (137.327)		Hf Hafnium (178.49)	Ta Tantalum (180.948)	W Tungsten (183.84)	Re Rhenium (186.207)	Os Osmium (190.23)	Ir Iridium (192.222)	Pt Platinum (195.084)	Au Gold (196.967)	Hg Mercury (200.59)	Tl Thallium (204.383)	Pb Lead (207.2)	Bi Bismuth (208.980)	Po Polonium (209)	At Astatine (210)	Rn Radon (222)			
7	Fr Francium (223)	Ra Radium (226)		Rf Rutherfordium (261)	Db Dubnium (262)	Sg Seaborgium (263)	Bh Bohrium (264)	Hs Hassium (265)	Mt Meitnerium (266)	Uun Ununennium (267)	Uuu Ununennium (268)	Uub Unbinilium (269)									
			La Lanthanum (138.905)	Ce Cerium (140.12)	Pr Praseodymium (140.908)	Nd Neodymium (144.24)	Pm Promethium (145)	Sm Samarium (150.36)	Eu Europium (151.964)	Gd Gadolinium (157.25)	Tb Terbium (158.925)	Dy Dysprosium (162.50)	Ho Holmium (164.930)	Er Erbium (167.26)	Tm Thulium (168.935)	Yb Ytterbium (173.054)	Lu Lutetium (174.967)				
			Ac Actinium (227)	Th Thorium (232.038)	Pa Protactinium (231.036)	U Uranium (238.029)	Np Neptunium (237)	Pu Plutonium (244)	Am Americium (243)	Cm Curium (247)	Bk Berkelium (247)	Cf Californium (251)	Es Einsteinium (252)	Fm Fermium (257)	Md Mendelevium (258)	No Nobelium (259)	Lr Lawrencium (260)				



Units and conversion

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Energy units conversion				
	kWh	MJ	kcal	eV
kWh	1	3.6	8.60×10^2	2.25×10^{25}
MJ	0.278	1	2.39×10^2	6.24×10^{24}
kcal	1.16×10^{-3}	4.18×10^{-3}	1	2.61×10^{22}
eV	4.45×10^{-26}	1.60×10^{-25}	3.83×10^{-23}	1

Time units conversion				
	s	h	d	y
s	1	2.78×10^{-4}	1.16×10^{-5}	3.16×10^{-8}
h	3.60×10^3	1	4.17×10^{-2}	1.14×10^{-4}
d	8.64×10^4	24	1	2.74×10^{-3}
y	3.16×10^7	8.77×10^3	3.65×10^2	1

Prefixes				
Z	zetta	10^{21}		<i>triljard</i>
E	exa	10^{18}		<i>triljoen</i>
P	peta	10^{15}	quadrillion	<i>biljard</i>
T	tera	10^{12}	trillion	<i>biljoen</i>
G	giga	10^9	billion	<i>miljard</i>
M	mega	10^6	million	<i>miljoen</i>

Other units conversion tables in book Ashby (inside back cover)	
Stress, pressure	
Power	



Energy usage and population

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Energy usage per person				
	World 2005	USA 2005	EU 2005	Netherlands 2006
Ratio	0.38	1.79	0.80	1
Joules	74 GJ/y 203 MJ/d	350 GJ/y 961 MJ/d	156 GJ/y 428 MJ/d	196 GJ/y 538 MJ/d
Liters oil	5 l/d	25 l/d	11 l/d	14 l/d
Continuous kW	2.3 kW	11.1 kW	4.9 kW	6.2 kW
In terms of food				55 x
Fossil	81 %	86 %	79 %	95 %
Nuclear	5 %	8 %	14 %	4 %
Renewable	14 %	7 %	7 %	1 %

Population (2009)	
World	6.79×10^9
China	1.34×10^9
EU	5.00×10^8
USA	3.07×10^8
Netherlands	1.67×10^7
Delft	9.62×10^4



Energy production and conversion

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Energy and CO ₂ production of fossil fuels				
Fuel	Spec. mass (kg/l)	kg OE	Energy	CO ₂
Coal, lignite		0.45	20 MJ/kg	1.6 kg/kg
Coal, anthracite		0.72	32 MJ/kg	2.9 kg/kg
Crude oil, diesel	0.86	1	38 MJ/l	3.1 kg/l
Gasoline	0.78	1.05	35 MJ/l	2.9 kg/l
Ethanol	0.74	0.71	23 MJ/l	2.8 kg/l
Liquid nat. gas	0.45	1.2	25 MJ/l	3.0 kg/l

Conversion efficiency (%)	
Mechanical → Electrical	80-98
Heat → Mechanical (steam engine)	40-55
Heat → Mechanical (diesel engine)	35
Heat → Mechanical (gasoline eng.)	20-25
Heat → Electrical (in power plant)	40
Heat → Electrical (from wall outlet)	35
Heat → Heat (kitchen gas burner)	50
Solar radiation → Electrical	12-17
Solar radiation → Biomass	0,2-5
Electrical → Mechanical	80-98
Electrical → Chemical (adv. battery)	85-90
Electrical → Visible light (inc. lamp)	15-20
Food → Mechanical (muscle power)	20-25



Data 2009

Everyday energy costs

Introduction, page 30

Primary power consumption		
	kW	l oil/h m ³ gas/h
Electric razor	0.03	0.003
Person	0.1	0.01
Candle	0.1	0.01
Laptop	0.3	0.03
Bicycle (extra food)	0.4	0.04
Kitchen gas burner	2	0.2
Vacuum cleaner	3	0.3
Full train per seat	8	0.8
Hot shower	10	1
Medium CV heater	20	2
Full car per seat	20	2
Full plane per seat	350	35

Energy cost of production		
	MJ	l oil
1 kg Al (recycled)	7	0.2
1 kg steel (from ore)	15	0.4
1 l milk	16	0.4
1 kg glass	18	0.5
1 kg paper	25	0.6
1 kg cheese	50	1.2
1 kg Al (from ore)	140	0.4
1 kg coffee	300	7
Passenger car	2 x 10 ⁴	500

Data 2009



How to produce 1000 MW electricity

Introduction, page 31

Resources needed to produce 1000 MW electrical energy (= 10% of NL electricity usage)	
Power plant, coal	3×10^9 kg coal/y
Power plant, oil	2×10^9 l oil/y
Power plant, natural gas	3×10^9 m ³ gas/y
Power plant, nuclear fission	1×10^8 kg U ore/y = 700 kg ²³⁵ U/y
Power plant, nuclear fusion	100 kg deuterium/y + 150 kg tritium/y
Wind turbines 90 m diameter, offshore	1300 turbines = 600 km ²
Solar cells (15 % efficiency)	60 km ²
Plants and trees	3000 km ²
People on home trainers (8 hours/day)	30 million people

Data 2009



50 l gasoline and alternatives

Introduction, page 32

Passenger car alternatives for 50 l gasoline		
	Volume (l) (tank included)	Mass (kg) (tank included)
Gasoline	52	60
H ₂ , high pressure	300	400
H ₂ , liquid	100	150
H ₂ , in LaNi ₅ H ₆	200	500
Flywheel	400	1000
Lead batteries	1500	3000

Data 2009

