EE1320: Measurement Science

Lecture 1: Introduction to Measurement and Measurement Systems

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Mobile measurement system





Measurement science is everywhere!



The importance of measurements

- The numbers tell the tale!
- For scientists, measuring is the way to test a theory
- For engineers, measuring is the way to validate a design
- Measurement systems and sensors are the senses of the computer
- Measuring also implies: knowing what you *don't* know



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"We have lots of information technology. We just don't have any information."



Course program 2013

week	date	topic
4.1	Tu 23/4	#1 intro measurements and meas. systems
	Fr 26/4	#2 sensors
4.3	Tu 7/5	#3 sensor readout and signal conditioning
4.4	Tu 14/5	#4 instrumentation amplifiers
	We 15/5	intermediate test
4.5	Tu 21/5	#5 analog-to-digital converters
4.6	We 29/5	#6 measurement instruments I
4.7	Tu 4/6	#7 measurement instruments II
	We 5/6	intermediate test
4.8	Tu 11/6	tutorial
4.11	We 3/7	final exam

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Learning objectives of this course

- Analyzing **measurement problems**:
 - you'll be able to
 - identify and describe a measurement problem
 - translate a measurement problem to measurable quantities
 - estimate whether a quantity is measurable under certain conditions
- Analyzing and interpreting **measurement results**:
 - you'll be able to
 - identify and describe sources of error



Learning objectives of this course

- Realization of simple measurement set-ups:
 - you'll be able to
 - apply **sensors** to measure non-electric quantities
 - apply simple **signal processing circuits** for sensor read-out
- Skillful use of **measurement instruments**:
 - you'll be able to
 - describe the operating principle of common instruments for electrical measurements
 - compare available instruments based on quality and accuracy



Overview course material

Book: "Regtien"

P.P.L Regtien, "Electronic Instrumentation" can be bought at VSSD (<u>http://www.vssd.nl</u>)

Reader: "Bell"

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S.A. Bell, "A beginner's guide to uncertainty in measurement"

Bell 1, Regtien 1.1

pdf is available on Blackboard or can be downloaded from http://www.npl.co.uk/publications/guides/





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Overview course material

Also on Blackboard:

- lecture slides
- overview exam material (what's covered, what's skipped?)
- overview course program
- study tips
- exercises, old exams with answers



Assessment

Final exam: week 4.11, Wednesday 3/7, 9:00 – 12:00 You can use a calculator and the book (Regtien)

Intermediate tests: check your understanding and earn a bonus of at most 1 point

week	date	test covers
4.4	We 15/5	lecture 1-3
4.7	We 5/6	lecture 1-6

The tests only count if the grade is at least a 6!

$$final\ grade = \min(10, exam + 0.1\frac{test1 + test2}{2})$$





Link with Semester Project EPO-2

- In EPO-2, you're developing a mine-detecting robot; you'll need measurement techniques to do so!
- EPO-2 'Just-in-time' training sessions related to EE1320:
 - week 3.1: Measuring with the robot
 - week 3.6: Opamps
 - week 4.2: Capacitive sensors
 - week 4.3: Inductive sensors



Today: introduction to measurement and measurement systems

- What exactly is measuring?
- How do you model a measurement system?
- Specification of the transfer of measurement systems

• Regtien: chapter 1, appendix A.1 Bell: sections 1, 2, 4 and 8





- quantity = property of a phenomenon or object that can be qualitatively distinguished, and <u>quantitatively</u> determined
 - length, time, mass, temperature, electrical resistance



Standards

• Objective, comparable measurements require **standards**



NIST-F1 time- and frequency standards: 1s = 9.191.631.770 periods of the resonance frequency of a Cs-133 atom



Copy of the *international prototype kilogram* (cylinder of 90% platinum and 10% iridium)



International System of Units (SI)



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Regtien A.1

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Kilograms vs. pounds: "The Gimli Glider"

http://www.youtube.com/watch?v=4yvUi7OAOL4

- 1983: Canada switches from imperial units to metric
- Air Canada flight 143: mistake with the fuel transfer...
 - 22.300 kg needed for the flight
 - 7.682 I left in the tank
 - -- how many kg is that?
 - the crew use 1.77 lb/l instead of 0.8 kg/l
 - consequence: only 4916 l transferred in stead of the required 20088 l
 - emergency landing!!



July 23, 1983: Air Canada flight 143 lands at a closed air force base in Gimli, Canada



Measurement Uncertainty



- Every measurement y of a quantity x is subject to measurement uncertainty
- Many causes:
 - random variations in the measurement value
 - varying measurement conditions
 - finite resolution / incorrect reading
 - deviations in the transfer of the measurement system
 - poorly defined definition of the quantity to be measured



• ...

Uncertainty vs. error

- Measurement uncertainty ≠ measurement error
 - Error: difference between measured value and the 'true' value
 - Uncertainty: quantification of the doubt about the measurement
- Measurement uncertainty can be quantified by
 - a standard deviation:
 "the mass is 100.02147 g with a standard deviation of 0.35 mg"
 - a confidence interval:

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"the mass is (100.02147 \pm 0.00079) g, at a confidence level of 95%"

- Unknown measurement errors contribute to the measurement uncertainty
- Some measurement errors can be determined, by means of calibration, and be corrected for afterwards.



Calibration

- Calibration makes a connection between
 - measurement values produced by a measurement instrument
 - corresponding values realized by standards
- Calibration procedure: comparison of an instrument with a (more accurate) measurement standard
- Calibration enables measurements which are traceable to standards
 - through an unbroken chain of comparisons
 - with associated specified uncertainties !





Example: calibration of a platinum thermometer

- PRT: platinum resistance thermometer
- typical **transfer**:



Example: calibration of a platinum thermometer

- Calibration procedure:
 - comparison to a more accurate reference thermometer (the working standard) at various calibration temperatures
 - \Rightarrow list of measured temperatures and resistance values with measurement uncertainty
 - determination of the coefficients of a formula that relates measured resistance to temperature

 $\Rightarrow R(t) = R_0 (1 + A \cdot t + B \cdot t^2)$

- determination of the corresponding measurement uncertainty
- Next, when using the thermometer, this formula will be used to translate a measured resistance into temperature



Structure of a measurement system







Structure of a measurement system



Transfer of a measurement system



• Using the transfer function H, output signal (indication) y_1 can be translated back to a measurement value x_1



Sensitivity



• **Sensitivity**: $S = \Delta y / \Delta x$

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• Ideal linear transfer: sensitivity S = y/x

Differential Sensitivity



Sensitivity example: weighing scale





of the measurement system

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Measurement errors due to deviations in the transfer



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- x_1 actual value of quantity to be measured
- *y*₁ output signal of measurement system
- *x*'₁ measured value determined based on nominal transfer

x₁ measurement error

Deviation in linear transfer



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Non-linearity

- A non-linear transfer will give measurement errors if the nominal transfer is assumed to be linear
- Integral non-linearity:



Ambiguity

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 No one-to-one relation between the quantity to be measured x and the output signal y



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Ambiguity

• Moving coil meter with friction







saturation





Ambiguity

saturation hysteresis dead zone



Opamp





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Cross-sensitivity

• Unwanted sensitivity to an **influence quantity** *c*





• Temperature sensitivity resistors typically expressed as temperature coefficient: $\alpha = \frac{1}{R} \cdot \frac{dR}{dT} [K^{-1}] \implies \frac{\Delta R}{R} = \alpha \cdot \Delta T$

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• Cross-sensitivity:
$$S_{temp} = \frac{\Delta U_o}{\Delta T} = \frac{\alpha \cdot U_g}{2}$$

• Let: $\alpha = 2,0 \cdot 10^{-5} \text{ K}^{-1}, U_g = 10 \text{ V} \Rightarrow S_{temp} = 0,10 \text{ mV/K}$ - 62 uV/ kgerror: 16 kg/ K !!

$$S_{massa} = 6,2 \ \mu V / kg$$





• **Compensation** for temperature differences: 4 strain gauges mounted to the base \Rightarrow small ΔT

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Measurement errors due to aging

- Measurement systems change over time
 - \Rightarrow uncertainty increases
 - \Rightarrow frequent calibration is needed



Accuracy Specifications ± (% of reading + % of range)

Function	Range ³	24 Hour² 23°C ±1°C	90 Day 23°C ±5°C	1 Year 23°C ±5°C
DC voltage	100.0000 mV	0.0030 + 0.0030	0.0040 + 0.0035	0.0050 + 0.0035
	1.000000 V	0.0020 + 0.0006	0.0030 + 0.0007	0.0040 + 0.0007
	10.00000 V	0.0015 + 0.0004	0.0020 + 0.0005	0.0035 + 0.0005
	100.0000 V	0.0020 + 0.0006	0.0035 + 0.0006	0.0045 + 0.0006
	1000.000 V	0.0020 + 0.0006	0.0035 + 0.0010	0.0045 + 0.0010



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Resolution



- **Resolution:** smallest change in x that causes a noticeable change in y
- Expressed in two ways:
 - absolute: Δx
 - relative to the full scale: $\Delta x / x_{max}$ often in bits: $-2\log(\Delta x / x_{max})$
- Example: 41/2-digit display, 200V range
 - $\Delta x = 0.01 \text{ V}$
 - $\Delta x / x_{max} = 0.01 \text{ V} / 199.99 \text{ V} = 5 \cdot 10^{-5}$ in bits: $-2\log (5 \cdot 10^{-5}) = 14.3 \text{ bits}$



Resolution examples





Resolution examples

• Resolution determined by display:







Summary

- Measuring = determining the value of a quantity
 - measurement requires international standards
 - calibration is needed for traceable, comparable measurements
 - every measurement is subject to measurement uncertainty
- Measurement system: converts quantity to be measured x into usable output signal y (often electrical, digital)
 - Data acquisition: $x \rightarrow \text{sensor} \rightarrow \text{signal conditioning} \rightarrow \text{ADC} \rightarrow y$
 - Characterized by transfer y = H(x) with sensitivity H'(x)
 - Deviations in the transfer can lead to measurement errors: non-linearity, ambiguity, cross sensitivity, finite resolution



What's next?

- Study:
 - Regtien chapter 1 and appendix A.1
 - Bell sections 1, 2, 4 en 8
- Practice:
 - Look at the exercises on Blackboard!
- Questions, things unclear? Let me know! <u>M.A.P.Pertijs@tudelft.nl</u>

Next time: Sensors

