

Solution for Exercise-sheet 3

I. NORMALIZATION OF SCHRÖDINGER'S WAVE FUNCTION

(I.1) (a) Normalize the wave function to determine the constant A.

Solution

$$\int_0^a |\psi(x)|^2 dx = 1 = |A|^2 \int_0^a (\sin \frac{n\pi}{a} x)^2 dx \quad (1)$$

$$= |A|^2 \int_0^a \sin^2 \frac{n\pi}{a} x dx \quad (2)$$

$$= |A|^2 \int_0^a \frac{1 - \cos \frac{2n\pi}{a} x}{2} dx \quad (3)$$

$$= |A|^2 \left[\frac{1}{2} x - \frac{1}{4n\pi} \sin(\frac{2n\pi}{a} x) \right]_0^a \quad (4)$$

$$= |A|^2 \times \frac{a}{2} \quad (5)$$

This yields

$$|A| = \sqrt{\frac{2}{a}}. \quad (6)$$

Therefore $A = \pm \sqrt{\frac{2}{a}}, \pm j \sqrt{\frac{2}{a}}$ or any complex number having the magnitude of $\sqrt{\frac{2}{a}}$

(b) What would be the time-dependent wave function?

Solution:

$$\Psi(x, t) = \psi(x) e^{-i \frac{E}{\hbar} t} = \sqrt{\frac{2}{a}} \sin \frac{n\pi x}{a} e^{-i \frac{E}{\hbar} t} = \sqrt{\frac{2}{a}} \frac{\{e^{i(\frac{n\pi}{a} x - \frac{E}{\hbar} t)} - e^{-i(\frac{n\pi}{a} x + \frac{E}{\hbar} t)}\}}{2i} \quad (7)$$

II. TIME-INDEPENDENT SCHRÖDINGER EQUATION

(II.1)

(a) Calculate the first two energy levels that the electron may occupy.

Solution: $E_n = \frac{\hbar^2 n^2 \pi^2}{2ma^2} = n^2 4.179 \times 10^{-20} \text{J} = 0.26n^2 \text{eV}$

The 1st energy level $n = 1, E_1 = 0.261 \text{eV}$

The 2nd energy level $n = 2, E_2 = 1.04 \text{eV}$

(b) If an electron drops from the second energy level to the first, what is the wavelength of the photon that might be emitted?

Solution: $\lambda = \frac{hc}{E_2 - E_1} = 1.59 \times 10^{-6} \text{m} = 1.59 \mu\text{m}$

(II.2) Consider the particle in the infinite potential well as shown in Figure.1.

(a) Using a boundary condition, derive an expression of the wave functions.

(b) Using the normalization, determine a constant of the amplitude of the wave function. The following formula could be used;

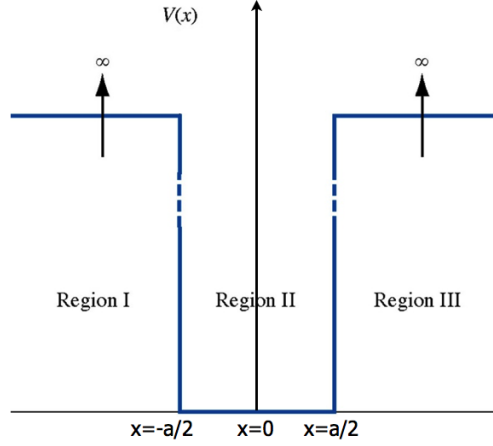


Figure 1: Potential function

$$\cos^2 x = \frac{\cos 2x + 1}{2} \quad (8)$$

III. TIME-INDEPENDENT SCHRÖDINGER EQUATION

(III.1) For the 1D step potential function shown in Figure 1, assume that the total energy of electrons $E > V_0$ and that the electrons are incident from the $+x$ direction traveling in the $-x$ direction.

(a) Write the general solutions of time-independent Schrödinger's equation for each region.

(b) By applying boundary conditions, derive expressions for the transmission T and reflection coefficients R , respectively.

Solution:

(a) For region II, $x > 0$

$$\frac{\partial^2 \psi_2(x)}{\partial x^2} + \frac{2m(E - V_0)}{\hbar^2} \psi_2(x) = 0. \quad (9)$$

General form of the solution is

$$\psi_2(x) = A_2 e^{jK_2 x} + B_2 e^{-jK_2 x} \quad (10)$$

where

$$K_2 = \sqrt{\frac{2m}{\hbar^2} (E - V_0)} \quad (11)$$

Term with B_2 represents incident wave (\leftarrow), and term with A_2 represents the reflected wave (\rightarrow).

Region I, $x < 0$

$$\frac{\partial^2 \psi_1(x)}{\partial x^2} + \frac{2m(E)}{\hbar^2} \psi_1(x) = 0. \quad (12)$$

The general solution is of the form

$$\psi_1(x) = A_1 e^{jK_1 x} + B_1 e^{-jK_1 x} \quad (13)$$

where

$$K_1 = \sqrt{\frac{2mE}{\hbar^2}} \quad (14)$$

The term involving B_1 represents the transmitted wave (\leftarrow). If a particle is transmitted into region I, it will not be reflected so that

$$A_1 = 0. \quad (15)$$

Then

$$\psi_1(x) = B_1 e^{-jK_1 x} \quad (16)$$

$$\psi_2(x) = A_2 e^{jK_2 x} + B_2 e^{-jK_2 x} \quad (17)$$

(b) Boundary conditions:

$$(1) \quad \psi_1(x=0) = \psi_2(x=0)$$

$$(2) \quad \left. \frac{\partial \psi_1(x)}{\partial x} \right|_{x=0} = \left. \frac{\partial \psi_2(x)}{\partial x} \right|_{x=0} \quad (18)$$

Applying the boundary conditions to the solutions, we find

$$B_1 = A_2 + B_2 \quad (19)$$

$$K_2 A_2 - K_2 B_2 = -K_1 B_1 \quad (20)$$

Combining these two equations, we find

$$A_2 = \left(\frac{K_2 - K_1}{K_2 + K_1} B_2 \right) \quad (21)$$

$$B_1 = \left(\frac{2K_2}{K_2 + K_1} B_2 \right) \quad (22)$$

The reflection coefficient is

$$R = \frac{A_2 A_2^*}{B_2 B_2^*} \rightarrow R = \frac{\left(\frac{K_2 - K_1}{K_1 + K_2} \right)^2}{1} \quad (23)$$

The transmission coefficient is

$$T = 1 - R \rightarrow T = \frac{4K_1 K_2}{(K_1 + K_2)^2} \quad (24)$$